

IN THE CLAIMS:

Claims 1-16 (Cancelled)

17. (Previously Presented) A fuel comprising:  
a source of at least one hydrido hydride ion;  
a source of protons.
18. (Previously Presented) A fuel according to claim 17, further comprising at least one burn modulator.
19. (Previously Presented) A fuel according to claim 17, wherein said fuel is formulated for use as (Previously Presented) A rocket fuel.
20. (Previously Presented) A fuel according to claim 17, wherein said fuel is liquid at ambient temperature and pressure.
21. (Previously Presented) A fuel according to claim 17, wherein said fuel is gaseous at ambient temperature and pressure.
22. (Previously Presented) A fuel according to claim 17, wherein said fuel is solid at ambient temperature and pressure.
23. (Previously Presented) A fuel according to claim 17, wherein said source of said protons comprises an acid.
24. (Previously Presented) A fuel according to claim 23, wherein said acid is a super-acid.

25. (Previously Presented) A fuel according to claim 23, wherein said acid is selected from the group consisting of HF, HCl, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>, the reaction products of HF and SbF<sub>5</sub>, the reaction products of HCl and Al<sub>2</sub>Cl<sub>6</sub>, the reaction products of H<sub>2</sub>SO<sub>3</sub>F and SbF<sub>5</sub>, the reaction products of H<sub>2</sub>SO<sub>4</sub> and SO<sub>2</sub>, and combinations thereof.
26. (Previously Presented) A fuel according to claim 17, wherein said source of protons comprises H<sup>1</sup>.
27. (Previously Presented) A fuel according to claim 17, wherein said source of protons comprises H<sup>2</sup>.
28. (Previously Presented) A fuel according to claim 17, wherein said source of protons comprises H<sup>3</sup>.
29. (Previously Presented) A fuel according to claim 17, wherein said source of hydride ion comprises at least one compound comprising a hydrino hydride ion and at least one other element.
30. (Previously Presented) A fuel according to claim 29, wherein said compound comprises at least one hydrino atom having a binding energy of about  $13.6/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer greater than 1 and at least one other element.
31. (Previously Presented) A fuel according to claim 29, wherein said compound comprises at least one dihydrino molecule having a first binding energy of about  $15.5/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer greater than 1 and at least one other element.

32. (Previously Presented) A fuel according to claim 29, wherein said compound comprises at least one dihydrino molecular ion having a first binding energy of about  $16.4/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer greater than 1, and at least one other element.
33. (Previously Presented) A fuel according to claim 29, wherein said compound comprises a hydrino hydride ion having a binding energy of about 0.65 eV and at least one other element.
34. (Previously Presented) A fuel according to claim 29, wherein the compound further comprises one or more selected from the group consisting of ordinary hydrogen molecules, ordinary hydride ions, ordinary hydrogen atoms, protons, ordinary hydrogen molecular ions, and ordinary  $H^{3+}$  ions.
35. (Previously Presented) A fuel according to claim 29, wherein the compound has a formula selected from the group of formulae consisting of  $MH$ ,  $MH_2$ , and  $M_2H_2$  wherein M is an alkali cation and H is selected from the group consisting of increased binding energy hydride ions, hydrino atoms and dihydrino molecules.
36. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MH_n$  wherein n is 1 or 2, M is an alkaline earth cation and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
37. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MHX$  wherein M is an alkali cation, X is one of a neutral atom, a molecule, or a singly negatively charged anion, and H is elected from the group consisting of hydrino hydride ions and hydrino atoms.

38. (Previously Presented) A fuel according to claim 37, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
39. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula MHX wherein M is an alkaline earth cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
40. (Previously Presented) A fuel according to claim 39, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
41. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula MHX wherein M is an alkaline earth cation, X is a doubly negatively charged anion, and H is a hydrino atom.
42. (Previously Presented) A fuel according to claim 41, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
43. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2HX$  wherein M is an alkali cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
44. (Previously Presented) A fuel according to claim 43, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.

45. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MH_n$  wherein  $n$  is an integer from 1 to 5,  $M$  is an alkali cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
46. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2H_n$  wherein  $n$  is an integer from 1 to 4,  $M$  is an alkaline earth cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
47. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2XH_n$  wherein  $n$  is an integer from 1 to 3,  $M$  is an alkaline earth cation,  $X$  is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
48. (Previously Presented) A fuel according to claim 47, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
49. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2X_2H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkaline earth cation,  $X$  is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
50. (Previously Presented) A fuel according to claim 49, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
51. (Previously Presented) A fuel according to claim 29, wherein the compound has the

formula  $M_2X_3H$  wherein M is an alkaline earth cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.

52. (Previously Presented) A fuel according to claim 51, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
53. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2XH_n$  wherein n is 1 or 2, M is an alkaline earth cation, X is a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
54. (Previously Presented) A fuel according to claim 53, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
55. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2XX'H$  wherein M is an alkaline earth cation, X is a singly negatively charged anion, X' is a doubly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
56. (Previously Presented) A fuel according to claim 55, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
57. (Previously Presented) A fuel according to claim 55, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.

58. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MM'H_n$  wherein  $n$  is an integer from 1 to 3,  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
59. (Previously Presented) A fuel according to claim 29, wherein the compound is  $MM'XH_n$  wherein  $n$  is 1 to 2,  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
60. (Previously Presented) A fuel according to claim 59, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
61. (Previously Presented) A fuel according to claim 29, wherein the compound is  $MM'XH$  where  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  is a doubly negatively charged anion, and  $H$  is selected from the group consisting of hydride ions and hydride atoms.
62. (Previously Presented) A fuel according to claim 61, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
63. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MM'XX'H$  where  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  and  $X'$  are each a singly negatively charged anion, and  $H$  is selected from the group consisting of hydride ions and hydride atoms.
64. (Previously Presented) A fuel according to claim 63, wherein said singly negatively

charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.

65. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $H_nS$  wherein  $n$  is 1 or 2, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species.
66. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MSiH_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species.
67. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MXM'H_n$  wherein  $n$  is an integer from 1 to 5;  
     $M$  is an alkali or alkaline earth cation;  
     $X$  is a singly negatively charged anion or a doubly negatively charged anion;  
     $M'$  is selected from the group consisting of Si, Al, Ni, the transition elements, the inner transition elements, and the rare earth elements; and  
    the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
68. (Previously Presented) A fuel according to claim 67, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
69. (Previously Presented) A fuel according to claim 67, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.



70. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MAIH_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
71. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MH_n$  wherein:
- $n$  is an integer from 1 to 6;
  - $M$  is selected from the group consisting of the transition elements, the inner transition elements, and the rare earth element cations and nickel; and
  - the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
72. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MNiH_n$  wherein:
- $n$  is an integer from 1 to 6;
  - $M$  is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum; and
  - the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
73. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MM'H_n$  wherein:
- $n$  is an integer from 1 to 6;
  - $M$  is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum;
  - $M'$  is selected from the group consisting of the transition elements, the inner transition elements, and rare earth element cations; and

the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

74. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2SiH_n$  wherein  $n$  is an integer from 1 to 8,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
75. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $Si_2H_n$  wherein  $n$  is an integer from 1 to 8, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
76. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $SiH_n$  wherein  $n$  is an integer from 1 to 8, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
77. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $TiH_n$  wherein  $n$  is an integer from 1 to 4, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
78. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $Al_2H_n$  wherein  $n$  is an integer from 1 to 4 and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
79. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MXAIX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy

hydrogen species.

80. (Previously Presented) A fuel according to claim 79, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
81. (Previously Presented) A fuel according to claim 79, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
82. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MXSiX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
83. (Previously Presented) A fuel according to claim 82, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
84. (Previously Presented) A fuel according to claim 82, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
85. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $SiO_2H_n$  wherein  $n$  is an integer from 1 to 6 and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
86. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MSiO_2H_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth

cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

87. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $MSi_2H_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
88. (Previously Presented) A fuel according to claim 29, wherein the compound has the formula  $M_2SiH_n$  wherein  $n$  is an integer from 1 to 8,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
89. (Previously Presented) A fuel according to claim 29, wherein the compound is greater than 50 atomic percent pure.
90. (Previously Presented) A fuel according to claim 29, wherein the compound is greater than 90 atomic percent pure.
91. (Previously Presented) A fuel according to claim 29, wherein said at least one other element comprises at least one selected from the group consisting of a proton, ordinary hydrogen atom, ordinary hydride ions, ordinary hydrogen molecules, ordinary hydrogen molecular ions and ordinary  $H_3^+$  ion .
92. (Previously Presented) A fuel according to claim 29, wherein said at least one other element comprises at least one element selected from the group consisting of alkaline earth metals and alkali metals .

93. (Previously Presented) A fuel according to claim 92, wherein said element comprises lithium or lithium ion.
94. (Previously Presented) A fuel according to claim 29, wherein said at least one other element comprises at least one element selected from the group consisting of organic compounds .
95. (Previously Presented) A fuel according to claim 29, wherein said at least one other element comprises at least one element selected from the group consisting of semiconductors .
96. (Previously Presented) A fuel according to claim 29, wherein said compound comprising:
- (a) at least one neutral, positive or negative increased binding energy hydrogen species having a binding energy:
    - (i) greater than the binding energy of the corresponding ordinary hydrogen species, or
    - (ii) greater than the binding energy of any hydrogen species for which the corresponding ordinary hydrogen species is unstable or is not observed because the ordinary hydrogen species' binding energy is less than thermal energies at ambient conditions, or is negative; and
  - (b) at least one other element, wherein said increased binding energy hydrogen species is selected from the group consisting of  $H_n$ ,  $H_n^-$ , and  $H_n^+$ , where n is an integer of 1 to 8, and n is greater than 1 when H has a positive charge.
97. (Previously Presented) A fuel according to claim 96, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydride ion

having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where  $p$  is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge; (b) hydrogen atom having a binding energy greater than about 13.6 eV; (c) hydrogen molecule having a first binding energy greater than about 15.5 eV; and (d) molecular hydrogen ion having a binding energy greater than about 16.4 eV.

98. (Previously Presented) A fuel according to claim 96, wherein the increased binding energy hydrogen species comprises a hydride ion having a binding energy of about 3.0, 6.6, 11.2, 16.7, 22.8, 29.3, 36.1, 42.8, 49.4, 55.5, 61.0, 65.6, 69.2, 71.53, 72.4, 71.54, 68.8, 64.0, 56.8, 47.1, 34.6, or 19.2.
99. (Previously Presented) A fuel according to claim 96, wherein said increased binding energy hydrogen species is a hydride ion having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where  $p$  is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge.

100. (Previously Presented) A fuel according to claim 96, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydrino atom having a binding energy of about  $13.6 \text{ eV}/(1/p)^2$ , where  $p$  is an integer greater than 1; (b) a hydride ion having a binding energy represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi \mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where  $p$  is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge; (c) a trihydrino molecular ion,  $\text{H}_3^+(1/p)$ , having a binding energy of about  $22.6/(1/p)^2 \text{ eV}$ ; (d) a dihydrino molecule having a binding energy of about  $15.5/(1/p)^2 \text{ eV}$ ; and (e) a dihydrino molecular ion with a binding energy of about  $16.4/(1/p)^2 \text{ eV}$ .

101. (Previously Presented) A fuel according to claim 100, wherein  $p$  is 2 to 200.
102. (Previously Presented) A fuel according to claim 100, wherein  $p$  is 2 to 24.
103. (Previously Presented) A fuel according to claim 100, wherein said increased binding energy hydrogen species is negative.

104. (Previously Presented) A fuel according to claim 17, wherein said source of hydrino hydride ion comprises a source of 2 or more types of hydrino hydride ions.
105. (Previously Presented) A method of propelling an object comprising:  
reacting at least one hydrino hydride ion with protons to form dihydrino molecules in such a manner as to impart acceleration to an object.
106. (Previously Presented) A method according to claim 105, wherein said object is a rocket.
107. (Previously Presented) A method according to claim 105, wherein said object is a projectile.
108. (Previously Presented) A method according to claim 105, wherein said reaction is initiated by heat energy.
109. (Previously Presented) A method according to claim 105, wherein said protons are supplied by a source of said protons comprising an acid.
110. (Previously Presented) A method according to claim 109, wherein said acid is a super-acid.
111. (Previously Presented) A method according to claim 109, wherein said acid is selected from the group consisting of HF, HCl, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>, the reaction products of HF and SbF<sub>5</sub>, the reaction products of HCl and Al<sub>2</sub>Cl<sub>6</sub>, the reaction products of H<sub>2</sub>SO<sub>3</sub>F and SbF<sub>5</sub>, the reaction products of H<sub>2</sub>SO<sub>4</sub> and SO<sub>2</sub>, and combinations thereof.
112. (Previously Presented) A method according to claim 105, wherein said protons



comprise mainly  $H^1$ .

113. (Previously Presented) A method according to claim 105, wherein said protons comprise mainly  $H^2$ .
114. (Previously Presented) A method according to claim 105, wherein said protons comprise mainly  $H^3$ .
115. (Previously Presented) A method according to claim 105, wherein said reaction is initiated by a rapid mixing of said hydrino hydride ion with a source of said protons.
116. (Previously Presented) A method according to claim 115, wherein said source of protons comprises an acid.
117. (Previously Presented) A method according to claim 105, further comprising decomposing a source of said hydrino hydride ion to provide said at least one hydrino hydride ion, wherein said source of said hydrino hydride ion comprising at least one compound comprising said least one hydrino hydride ion and at least one other element.
118. (Previously Presented) A method according to claim 117, wherein said compound comprises at least one hydrino atom having a binding energy of about  $13.6/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer greater than 1 and at least one other element.
119. (Previously Presented) A method according to claim 117, wherein said compound comprises at least one dihydrino molecule having a first binding energy of about  $15.5/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer

greater than 1 and at least one other element.

120. (Previously Presented) A method according to claim 117, wherein said compound comprises at least one dihydrino molecular ion having a first binding energy of about  $16.4/n^2$  eV, wherein n is a fraction whose numerator is 1 and denominator is an integer greater than 1, and at least one other element.
121. (Previously Presented) A method according to claim 117, wherein said compound comprises a hydrino hydride ion having a binding energy of about 0.65 eV and at least one other element.
122. (Previously Presented) A method according to claim 117, wherein the compound further comprises one or more selected from the group consisting of ordinary hydrogen molecules, ordinary hydride ions, ordinary hydrogen atoms, protons, ordinary hydrogen molecular ions, and ordinary  $H^{3+}$  ions; and said method further comprises decomposing said compound to provide said hydrino hydride ion and protons.
123. (Previously Presented) A method according to claim 117, wherein the compound has a formula selected from the group of formulae consisting of MH,  $MH_2$ , and  $M_2H_2$  wherein M is an alkali cation and H is selected from the group consisting of increased binding energy hydride ions, hydrino atoms and dihydrino molecules, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
124. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MH_n$  wherein n is 1 or 2, M is an alkaline earth cation and H is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

125. (Previously Presented) A method according to claim 117, wherein the compound has the formula MHX wherein M is an alkali cation, X is one of a neutral atom, a molecule, or a singly negatively charged anion, and H is elected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
126. (Previously Presented) A method according to claim 125, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
127. (Previously Presented) A method according to claim 117, wherein the compound has the formula MHX wherein M is an alkaline earth cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
128. (Previously Presented) A method according to claim 127, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
129. (Previously Presented) A method according to claim 117, wherein the compound has the formula MHX wherein M is an alkaline earth cation, X is a doubly negatively charged anion, and H is a hydrino atom, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
130. (Currently Amended) A method ~~compound~~ according to claim 129, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.

131. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2HX$  wherein M is an alkali cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
132. (Previously Presented) A method according to claim 131, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
133. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MH_n$  wherein n is an integer from 1 to 5, M is an alkali cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
134. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2H_n$  wherein n is an integer from 1 to 4, M is an alkaline earth cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
135. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2XH_n$  wherein n is an integer from 1 to 3, M is an alkaline earth cation, X is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

136. (Previously Presented) A method according to claim 135, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
137. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2X_2H_n$  wherein n is 1 or 2, M is an alkaline earth cation, X is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
138. (Previously Presented) A method according to claim 137, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
139. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2X_3H$  wherein M is an alkaline earth cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
140. (Previously Presented) A method according to claim 139, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
141. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2XH_n$  wherein n is 1 or 2, M is an alkaline earth cation, X is a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one

increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

142. (Previously Presented) A method according to claim 141, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
143. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2XX'H$  wherein M is an alkaline earth cation, X is a singly negatively charged anion, X' is a doubly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
144. (Previously Presented) A method according to claim 143, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
145. (Previously Presented) A method according to claim 143, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
146. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MM'H_n$  wherein n is an integer from 1 to 3, M is an alkaline earth cation, M' is an alkali metal cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
147. (Previously Presented) A method according to claim 117, wherein the compound is

$MM'XH_n$  wherein  $n$  is 1 to 2,  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

148. (Previously Presented) A method according to claim 147, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
149. (Previously Presented) A method according to claim 117, wherein the compound is  $MM'XH$  where  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  is a doubly negatively charged anion, and  $H$  is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
150. (Previously Presented) A method according to claim 149, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
151. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MM'XX'H$  where  $M$  is an alkaline earth cation,  $M'$  is an alkali metal cation,  $X$  and  $X'$  are each a singly negatively charged anion, and  $H$  is selected from the group consisting of hydrino hydride ions and hydrino atoms, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
152. (Previously Presented) A method according to claim 151, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.

153. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $H_nS$  wherein  $n$  is 1 or 2, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
154. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MSiH_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
155. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MXM'H_n$  wherein  $n$  is an integer from 1 to 5;  
     $M$  is an alkali or alkaline earth cation;  
     $X$  is a singly negatively charged anion or a doubly negatively charged anion;  
     $M'$  is selected from the group consisting of Si, Al, Ni, the transition elements, the inner transition elements, and the rare earth elements; and  
    the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
156. (Previously Presented) A method according to claim 155, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
157. (Previously Presented) A method according to claim 155, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and



sulfate ions.

158. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MAIH_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
159. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MH_n$  wherein:
- $n$  is an integer from 1 to 6;
  - $M$  is selected from the group consisting of the transition elements, the inner transition elements, and the rare earth element cations and nickel; and
  - the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
160. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MNiH_n$  wherein:
- $n$  is an integer from 1 to 6;
  - $M$  is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum; and
  - the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
161. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MM'H_n$  wherein:

n is an integer from 1 to 6;

M is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum;

M' is selected from the group consisting of the transition elements, the inner transition elements, and rare earth element cations; and

the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

162. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $M_2SiH_n$  wherein n is an integer from 1 to 8, M is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
163. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $Si_2H_n$  wherein n is an integer from 1 to 8, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
164. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $SiH_n$  wherein n is an integer from 1 to 8, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
165. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $TiH_n$  wherein n is an integer from 1 to 4, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

166. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $Al_2H_n$  wherein  $n$  is an integer from 1 to 4 and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
167. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MXAlX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
168. (Previously Presented) A method according to claim 167, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
169. (Previously Presented) A method according to claim 167, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
170. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $MXSiX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

171. (Previously Presented) A method according to claim 170, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
172. (Previously Presented) A method according to claim 170, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
173. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $\text{SiO}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6 and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
174. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $\text{MSiO}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
175. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $\text{MSi}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.
176. (Previously Presented) A method according to claim 117, wherein the compound has the formula  $\text{M}_2\text{SiH}_n$  wherein  $n$  is an integer from 1 to 8,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy

hydrogen species, and said method further comprises decomposing said compound to provide said hydrino hydride ion.

177. (Previously Presented) A method according to claim 117, wherein the compound is greater than 50 atomic percent pure.
178. (Previously Presented) A method according to claim 117, wherein the compound is greater than 90 atomic percent pure.
179. (Previously Presented) A method according to claim 117, wherein said at least one other element comprises at least one selected from the group consisting of a proton, ordinary hydride ion, ordinary hydrogen atom, ordinary hydrogen molecules, ordinary hydrogen molecular ions and ordinary  $H_3^+$  ions and said method further comprises decomposing said compound to provide said hydrino hydride ion.
180. (Previously Presented) A method according to claim 117, wherein said at least one other element comprises at least one element selected from the group consisting of alkaline earth metals and alkali metals and said method further comprises decomposing said compound to provide said hydrino hydride ion.
181. (Previously Presented) A method according to claim 180, wherein said element comprises lithium or lithium ion.
182. (Previously Presented) A method according to claim 117, wherein said at least one other element comprises at least one element selected from the group consisting of organic compounds and said method further comprises decomposing said compound to provide said hydrino hydride ion.

183. (Previously Presented) A method according to claim 117, wherein said at least one other element comprises at least one element selected from the group consisting of semiconductors and said method further comprises decomposing said compound to provide said hydride ion.
184. (Previously Presented) A method according to claim 117, wherein said compound comprising:
- (a) at least one neutral, positive or negative increased binding energy hydrogen species having a binding energy:
    - (i) greater than the binding energy of the corresponding ordinary hydrogen species, or
    - (ii) greater than the binding energy of any hydrogen species for which the corresponding ordinary hydrogen species is unstable or is not observed because the ordinary hydrogen species' binding energy is less than thermal energies at ambient conditions, or is negative; and
  - (b) at least one other element, wherein said increased binding energy hydrogen species is selected from the group consisting of  $H_n$ ,  $H_n^-$ , and  $H_n^+$ , where n is an integer of 1 to 8, and n is greater than 1 when H has a positive charge.
185. (Previously Presented) A method according to claim 184, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydride ion having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where p is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and e is the elementary charge; (b) hydrogen atom having a binding energy greater than about 13.6 eV; (c) hydrogen molecule having a first binding energy greater than about 15.5 eV; and (d) molecular hydrogen ion having a binding energy greater than about 16.4 eV.

186. (Previously Presented) A method according to claim 184, wherein the increased binding energy hydrogen species comprises a hydride ion having a binding energy of about 3.0, 6.6, 11.2, 16.7, 22.8, 29.3, 36.1, 42.8, 49.4, 55.5, 61.0, 65.6, 69.2, 71.53, 72.4, 71.54, 68.8, 64.0, 56.8, 47.1, 34.6, or 19.2.

187. (Previously Presented) A method according to claim 184, wherein said increased binding energy hydrogen species is a hydride ion having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where p is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the

permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge.

188. (Previously Presented) A method according to claim 184, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydrino atom having a binding energy of about  $13.6 \text{ eV}/(1/p)^2$ , where  $p$  is an integer greater than 1; (b) a hydride ion having a binding energy represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_e e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where  $p$  is an integer greater than 1,  $s = 1/2$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge; (c) a trihydrino molecular ion,  $\text{H}_3^+ (1/p)$ , having a binding energy of about  $22.6/(1/p)^2 \text{ eV}$ ; (d) a dihydrino molecule having a binding energy of about  $15.5/(1/p)^2 \text{ eV}$ ; and (e) a dihydrino molecular ion with a binding energy of about  $16.4/(1/p)^2 \text{ eV}$ .

189. (Previously Presented) A method according to claim 188, wherein  $p$  is 2 to 200.
190. (Previously Presented) A method according to claim 188, wherein  $p$  is 2 to 24.
191. (Previously Presented) A method according to claim 188, wherein said increased binding energy hydrogen species is negative.
192. (Previously Presented) A method according to claim 105, wherein said at least one



hydrino hydride ion comprises a mixture of 2 or more types of hydrino hydride ions.

193. (Previously Presented) A method according to claim 105, further comprising reacting hydrino atoms with electrons to produce said hydrino hydride ion.
194. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst having a net enthalpy of reaction of at least  $m \cdot 27 \text{ eV}$ , where  $m$  is an integer.
195. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst adapted to provide a resonant absorption with the energy released by said hydrogen atoms when said hydrogen atoms undergo a transition to a lower energy state.
196. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising a salt of rubidium.
197. (Previously Presented) A method according to claim 196, wherein said salt of rubidium is selected from the group consisting of  $\text{RbOH}$ ,  $\text{Rb}_2\text{SO}_4$ ,  $\text{Rb}_2\text{CO}_3$ , and  $\text{Rb}_3\text{PO}_4$ .
198. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising a salt of potassium.
199. (Previously Presented) A method according to claim 198, wherein said salt of potassium is selected from the group consisting of  $\text{KOH}$ ,  $\text{K}_2\text{SO}_4$ ,  $\text{K}_2\text{CO}_3$  and  $\text{K}_3\text{PO}_4$ .

200. (Previously Presented) A method according to claim 198, wherein said salt of potassium is  $K_2CO_3$ .
201. (Previously Presented) A method according to claim 193, further comprising forming hydrino atoms from hydrogen atoms by use of a catalyst comprising a salt of titanium.
202. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising an ion selected from the group consisting of  $(Rb^+)$ ,  $(Mo^{2+})$ , and  $(Ti^{2+})$ .
203. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst selected from the group consisting of  $(Al^{2+})$ ,  $(Ar^+)$ ,  $(Ti^{2+})$ ,  $(As^{2+})$ ,  $(Rb^+)$ ,  $(Mo^{2+})$ ,  $(Ru^{2+})$ ,  $(In^{2+})$ , and  $(Te^{2+})$ .
204. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst capable of providing a net enthalpy of reaction in the range of 26.8 to 28.5 eV.
205. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising at least one pair of ions selected from the group consisting of:  $(Sn^{4+}, Si^{4+})$ ,  $(Pr^{3+}, Ca^{2+})$ ,  $(Sr^{2+}, Cr^{2+})$ ,  $(Cr^{3+}, Tb^{3+})$ ,  $(Sb^{3+}, Co^{2+})$ ,  $(Bi^{3+}, Ni^{2+})$ ,  $(Pd^{2+}, In^+)$ ,  $(La^{3+}, Dy^{3+})$ ,  $(La^{3+}, Ho^{3+})$ ,  $(K^+, K^+)$ ,  $(V^{3+}, Pd^{2+})$ ,  $(Lu^{3+}, Zn^{2+})$ ,  $(As^{3+}, Ho^{3+})$ ,  $(Mo^{5+}, Sn^{4+})$ ,  $(Sb^{3+}, Cd^{2+})$ ,  $(Ag^{2+}, Ag^+)$ ,  $(La^{3+}, Er^{3+})$ ,  $(V^{4+}, B^{3+})$ ,  $(Fe^{3+}, Ti^{3+})$ ,  $(Co^{2+}, Ti^+)$ ,  $(Bi^{3+}, Zn^{2+})$ ,  $(As^{3+}, Dy^{3+})$ ,  $(Ho^{3+}, Mg^{2+})$ ,  $(K^+, Rb^+)$ ,  $(Cr^{3+}, Pr^{3+})$ ,  $(Sr^{2+}, Fe^{2+})$ ,  $(Ni^{2+}, Cu^+)$ ,  $(Li^+, Pb^{2+})$ ,  $(Sr^{2+}, Mo^{2+})$ ,  $(Y^{3+}, Zr^{4+})$ ,  $(Cd^{2+}, Ba^{2+})$ ,  $(Ho^{3+}, Pb^{2+})$ ,  $(Eu^{3+}, Mg^{2+})$ ,  $(Er^{3+}, Mg^{2+})$ ,  $(Bi^{4+}, Al^{3+})$ ,  $(Ca^{2+}, Sm^{3+})$ ,  $(V^{3+}, La^{3+})$ ,  $(Gd^{3+}, Cr^{2+})$ ,  $(Mn^{2+}, Ti^+)$ ,  $(Yb^{3+}, Fe^{2+})$ ,  $(Ni^{2+}, Ag^+)$ ,  $(Zn^{2+}, Yb^{2+})$ ,  $(Se^{4+}, Sn^{4+})$ ,  $(Sb^{3+}, Bi^{2+})$ , and  $(Eu^{3+}, Pb^{2+})$ .

206. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising oxygen in combination with at least one atom selected from the group consisting of Cu, As, Pd, Te, Cs and Pt.
207. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising at least one pair selected from the group consisting of: (B, Li<sup>+</sup>), (S, Li<sup>+</sup>), (Br, Li<sup>+</sup>), (Pm<sup>+</sup>, Li<sup>+</sup>), (Sm<sup>+</sup>, Li<sup>+</sup>), (Tb<sup>+</sup>, Li<sup>+</sup>), (Dy<sup>+</sup>, Li<sup>+</sup>), (Sb<sup>+</sup>, H<sup>+</sup>) and (Bi<sup>+</sup>, H<sup>+</sup>).
208. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising at least one pair selected from the group consisting of:

( He 0+ , Co 3+ );	( O 1+ , Nd 4+ );	( Al 2+ , Cl 5+ );
( He 0+ , Ga 3+ );	( O 1+ , Tb 4+ );	( Al 4+ , Mn 8+ );
( Li 0+ , Ni 3+ );	( O 2+ , Ne 3+ );	( Si 1+ , Mg 2+ );
( Li 0+ , Xe 3+ );	( O 3+ , Sb 6+ );	( Si 1+ , V 2+ );
( Li 0+ , Hg 3+ );	( O 4+ , Fe 7+ );	( Si 1+ , Tc 2+ );
( Li 1+ , Na 4+ );	( F 0+ , Al 2+ );	( Si 1+ , Sn 2+ );
( Li 1+ , Y 6+ );	( F 0+ , Si 2+ );	( Si 1+ , Hf 2+ );
( Be 1+ , Bi 6+ );	( F 0+ , Fe 2+ );	( Si 1+ , Pb 2+ );
( Be 2+ , Al 6+ );	( F 0+ , Co 2+ );	( Si 2+ , Co 3+ );
( B 1+ , C 2+ );	( F 0+ , Ru 2+ );	( Si 2+ , Ga 3+ );
( B 1+ , K 2+ );	( F 0+ , In 2+ );	( Si 2+ , Ge 3+ );
( B 1+ , Ho 3+ );	( F 0+ , Sb 2+ );	( Si 2+ , Tl 3+ );
( B 1+ , Er 3+ );	( F 0+ , Bi 2+ );	( Si 3+ , Ni 6+ );
( B 1+ , Tm 3+ );	( F 1+ , Sb 4+ );	( Si 3+ , Rb 7+ );
( B 1+ , Lu 3+ );	( F 3+ , Fe 6+ );	( Si 4+ , Al 6+ );
( C 1+ , N 2+ );	( Ne 0+ , Sm 3+ );	( P 1+ , Mg 2+ );
( C 1+ , V 3+ );	( Ne 0+ , Dy 3+ );	( P 1+ , Tc 2+ );
( C 1+ , Tc 3+ );	( Ne 0+ , Ho 3+ );	( P 1+ , Sn 2+ );
( C 1+ , Ru 3+ );	( Ne 0+ , Er 3+ );	( P 1+ , Hf 2+ );
( C 1+ , Sn 3+ );	( Ne 0+ , Lu 3+ );	( P 1+ , Pb 2+ );
( C 2+ , Mn 4+ );	( Ne 1+ , N 3+ );	( P 2+ , Ni 3+ );
( C 2+ , Co 4+ );	( Ne 1+ , K 3+ );	( P 2+ , Cd 3+ );
( N 0+ , Sr 2+ );	( Ne 1+ , V 4+ );	( P 2+ , Xe 3+ );
( N 0+ , La 2+ );	( Ne 2+ , O 4+ );	( P 3+ , Nb 5+ );

( N 0+ , Ce 2+ );	(Na 0+ , Al 2+ );	(P 5+ , C 5+ );
( N 0+ , Pr 2+ );	(Na 0+ , Si 2+ );	(S 1+ , P 2+ );
( N 0+ , Nd 2+ );	(Na 0+ , Fe 2+ );	(S 1+ , Se 2+ );
( N 0+ , Pm 2+ );	(Na 0+ , Co 2+ );	(S 1+ , La 3+ );
( N 0+ , Sm 2+ );	(Na 0+ , Ru 2+ );	(S 1+ , Ce 3+ );
( N 0+ , Eu 2+ );	(Na 0+ , In 2+ );	(S 1+ , Au 2+ );
( N 1+ , O 2+ );	(Na 0+ , Sb 2+ );	(S 2+ , Sr 3+ );
( N 1+ , Si 3+ );	(Na 0+ , Bi 2+ );	(S 2+ , Cd 3+ );
( N 1+ , P 3+ );	(Na 2+ , Ti 5+ );	(S 3+ , Cu 4+ );
( N 1+ , Mn 3+ );	(Na 2+ , Kr 6+ );	(S 3+ , Rb 4+ );
( N 1+ , Rh 3+ );	(Na 3+ , Y 7+ );	(S 4+ , O 4+ );
( N 2+ , F 3+ );	(Mg 1+ , Rb 3+ );	(Cl 1+ , C 2+ );
( N 3+ , Br 6+ );	(Mg 1+ , Eu 4+ );	(Cl 1+ , K 2+ );
( O 0+ , Ti 2+ );	(Mg 3+ , Ne 5+ );	(Cl 1+ , Zr 3+ );
( O 0+ , V 2+ );	(Mg 6+ , Cl 8+ );	(Cl 1+ , Eu 3+ );
( O 0+ , Nb 2+ );	(Al 1+ , Sc 2+ );	(Cl 1+ , Tm 3+ );
( O 0+ , Hf 2+ );	(Al 1+ , Zr 2+ );	(Cl 2+ , Te 4+ );
( O 1+ , Ne 2+ );	(Al 1+ , Lu 2+ );	(Cl 2+ , Sm 4+ );
( O 1+ , Ca 3+ );	(Al 2+ , S 5+ );	(Cl 2+ , Gd 4+ );
( Cl 2+ , Ho 4+ );	(Sc 4+ , N 5+ );	(Mn 4+ , Ge 5+ );
( Cl 2+ , Er 4+ );	(Ti 2+ , Ar 2+ );	(Fe 1+ , Sc 2+ );
( Cl 3+ , Cl 4+ );	(Ti 2+ , Mo 3+ );	(Fe 1+ , Y 2+ );
( Cl 5+ , Ni 6+ );	(Ti 4+ , O 5+ );	(Fe 1+ , Yb 2+ );
( Cl 5+ , Cu 6+ );	(Ti 4+ , Zn 6+ );	(Fe 1+ , Lu 2+ );
( Cl 5+ , Rb 7+ );	(Ti 4+ , As 6+ );	(Fe 2+ , S 3+ );
( Ar 0+ , Ba 2+ );	(V 1+ , Sr 2+ );	(Fe 2+ , Cu 3+ );
( Ar 0+ , Ce 2+ );	(V 1+ , La 2+ );	(Fe 2+ , Zn 3+ );
( Ar 0+ , Pr 2+ );	(V 1+ , Ce 2+ );	(Fe 2+ , Br 3+ );
( Ar 0+ , Nd 2+ );	(V 1+ , Pr 2+ );	(Fe 2+ , Zr 4+ );
( Ar 0+ , Ra 2+ );	(V 1+ , Nd 2+ );	(Fe 2+ , Ce 4+ );
( Ar 1+ , Ti 3+ );	(V 1+ , Pm 2+ );	(Fe 5+ , Sr 7+ );
( Ar 2+ , C 3+ );	(V 1+ , Sm 2+ );	(Co 1+ , Mg 2+ );
( Ar 3+ , K 4+ );	(V 1+ , Eu 2+ );	(Co 1+ , Cr 2+ );
( Ar 3+ , Br 5+ );	(V 2+ , O 2+ );	(Co 1+ , Mn 2+ );
( Ar 3+ , Mo 5+ );	(V 3+ , Mn 4+ );	(Co 1+ , Mo 2+ );
( Ar 4+ , Y 5+ );	(V 3+ , Co 4+ );	(Co 1+ , Tc 2+ );
( K 1+ , Si 3+ );	(V 4+ , Ar 6+ );	(Co 1+ , Pb 2+ );
( K 1+ , P 3+ );	(V 4+ , Sc 5+ );	(Co 2+ , Cu 3+ );
( K 1+ , Mn 3+ );	(V 5+ , Mg 5+ );	(Co 2+ , Zn 3+ );
( K 1+ , Ge 3+ );	(V 6+ , Sc 8+ );	(Co 2+ , Br 3+ );
( K 1+ , Rh 3+ );	(V 6+ , Br 8+ );	(Co 2+ , Zr 4+ );
( K 1+ , Tl 3+ );	(Cr 1+ , Sc 2+ );	(Co 2+ , Ag 3+ );
( K 2+ , He 2+ );	(Cr 1+ , Ti 2+ );	(Co 2+ , Ce 4+ );
( K 2+ , Si 4+ );	(Cr 1+ , Zr 2+ );	(Co 2+ , Hf 4+ );
( K 2+ , As 4+ );	(Cr 1+ , Lu 2+ );	(Co 4+ , Nb 6+ );
( K 3+ , P 5+ );	(Cr 2+ , F 2+ );	(Co 5+ , Sc 6+ );
( K 3+ , Zr 5+ );	(Cr 2+ , Na 2+ );	(Ni 1+ , Co 2+ );
( K 4+ , Rb 6+ );	(Cr 2+ , Se 3+ );	(Ni 1+ , Ni 2+ );
( K 5+ , Mg 4+ );	(Cr 2+ , Pd 3+ );	(Ni 1+ , Rh 2+ );

( K 5+ , Kr 7+ );	(Cr 2+ , I 3+ );	(Ni 1+ , Cd 2+ );
( K 6+ , Y 8+ );	(Cr 2+ , Hg 3+ );	(Ni 1+ , Sb 2+ );
( Ca 1+ , C 2+ );	(Cr 3+ , O 3+ );	(Ni 2+ , Ne 2+ );
( Ca 1+ , Sm 3+ );	(Cr 3+ , Ni 4+ );	(Ni 2+ , Ca 3+ );
( Ca 1+ , Dy 3+ );	(Cr 4+ , O 4+ );	(Ni 2+ , Nd 4+ );
( Ca 1+ , Ho 3+ );	(Cr 5+ , Ne 5+ );	(Ni 2+ , Tb 4+ );
( Ca 1+ , Er 3+ );	(Cr 5+ , Fe 7+ );	(Ni 4+ , Rb 6+ );
( Ca 1+ , Tm 3+ );	(Mn 1+ , V 2+ );	(Ni 6+ , Ar 8+ );
( Ca 1+ , Lu 3+ );	(Mn 1+ , Nb 2+ );	(Cu 1+ , Ag 2+ );
( Ca 2+ , O 3+ );	(Mn 1+ , Sn 2+ );	(Cu 1+ , I 2+ );
( Ca 2+ , Ni 4+ );	(Mn 1+ , Hf 2+ );	(Cu 1+ , Cs 2+ );
( Ca 3+ , Mn 5+ );	(Mn 2+ , Cu 3+ );	(Cu 1+ , Au 2+ );
( Ca 3+ , Rb 5+ );	(Mn 2+ , Zn 3+ );	(Cu 1+ , Hg 2+ );
( Ca 4+ , Cl 6+ );	(Mn 2+ , Br 3+ );	(Cu 2+ , Sm 4+ );
( Ca 4+ , Ar 6+ );	(Mn 2+ , Zr 4+ );	(Cu 2+ , Gd 4+ );
( Ca 4+ , Sc 5+ );	(Mn 2+ , Ce 4+ );	(Cu 2+ , Dy 4+ );
( Ca 5+ , Y 7+ );	(Mn 2+ , Hf 4+ );	(Cu 3+ , K 4+ );
( Sc 2+ , Ti 4+ );	(Mn 3+ , Mg 3+ );	(Cu 3+ , Br 5+ );
( Sc 2+ , Bi 4+ );	(Mn 3+ , Te 5+ );	(Cu 3+ , Mo 5+ );
( Cu 4+ , Rb 6+ );	(Se 1+ , Fe 2+ );	(Sr 1+ , Ga 2+ );
( Cu 5+ , Mn 7+ );	(Se 1+ , Co 2+ );	(Sr 1+ , Te 2+ );
( Zn 1+ , P 2+ );	(Se 1+ , Ge 2+ );	(Sr 1+ , Pt 2+ );
( Zn 1+ , I 2+ );	(Se 1+ , Ru 2+ );	(Sr 1+ , Tl 2+ );
( Zn 1+ , La 3+ );	(Se 1+ , In 2+ );	(Sr 2+ , C 3+ );
( Zn 1+ , Au 2+ );	(Se 1+ , Bi 2+ );	(Sr 2+ , Mo 4+ );
( Zn 1+ , Hg 2+ );	(Se 2+ , Te 3+ );	(Sr 3+ , Ar 4+ );
( Zn 2+ , Ti 4+ );	(Se 3+ , Br 4+ );	(Sr 3+ , Sr 4+ );
( Zn 2+ , Sn 4+ );	(Se 5+ , Y 7+ );	(Sr 3+ , Sb 5+ );
( Zn 2+ , Bi 4+ );	(Br 1+ , P 2+ );	(Sr 3+ , Bi 5+ );
( Zn 3+ , As 5+ );	(Br 1+ , I 2+ );	(Sr 4+ , Ar 5+ );
( Zn 4+ , Sr 6+ );	(Br 1+ , La 3+ );	(Sr 4+ , Cu 5+ );
( Zn 5+ , Mn 7+ );	(Br 1+ , Au 2+ );	(Y 2+ , Sr 3+ );
( Zn 6+ , Mo 8+ );	(Br 3+ , He 2+ );	(Y 2+ , Cd 3+ );
( Ga 1+ , Cr 2+ );	(Br 3+ , Si 4+ );	(Y 3+ , Se 5+ );
( Ga 1+ , Mn 2+ );	(Br 3+ , Ge 4+ );	(Y 3+ , Pb 5+ );
( Ga 1+ , Fe 2+ );	(Br 4+ , S 5+ );	(Y 4+ , Ti 5+ );
( Ga 1+ , Ge 2+ );	(Br 4+ , Cl 5+ );	(Y 4+ , Zn 5+ );
( Ga 1+ , Mo 2+ );	(Br 5+ , Sb 6+ );	(Y 5+ , Co 6+ );
( Ga 1+ , Ru 2+ );	(Br 6+ , Ar 8+ );	(Y 6+ , K 7+ );
( Ga 1+ , Bi 2+ );	(Kr 1+ , B 2+ );	(Zr 2+ , P 2+ );
( Ga 2+ , Rb 3+ );	(Kr 1+ , S 2+ );	(Zr 2+ , Ag 2+ );
( Ga 2+ , Eu 4+ );	(Kr 1+ , Br 2+ );	(Zr 2+ , I 2+ );
( Ga 2+ , Tm 4+ );	(Kr 1+ , Xe 2+ );	(Zr 2+ , Cs 2+ );
( Ge 1+ , Mg 2+ );	(Kr 1+ , Nd 3+ );	(Zr 2+ , La 3+ );
( Ge 1+ , Mn 2+ );	(Kr 1+ , Pm 3+ );	(Zr 2+ , Au 2+ );
( Ge 1+ , Tc 2+ );	(Kr 1+ , Tb 3+ );	(Zr 2+ , Hg 2+ );
( Ge 1+ , Sn 2+ );	(Kr 2+ , Kr 3+ );	(Nb 2+ , C 2+ );
( Ge 1+ , Pb 2+ );	(Kr 2+ , Tb 4+ );	(Nb 2+ , K 2+ );
( Ge 2+ , F 2+ );	(Kr 3+ , O 3+ );	(Nb 2+ , Zr 3+ );

( Ge 2+ , Na 2+ );	(Kr 3+ , Ni 4+ );	(Nb 2+ , Eu 3+ );
( Ge 2+ , Se 3+ );	(Kr 3+ , Kr 4+ );	(Nb 2+ , Tm 3+ );
( Ge 2+ , Pd 3+ );	(Kr 3+ , Nb 5+ );	(Nb 2+ , Lu 3+ );
( Ge 2+ , I 3+ );	(Kr 4+ , Zr 5+ );	(Nb 3+ , Kr 3+ );
( Ge 3+ , V 5+ );	(Kr 5+ , Sr 6+ );	(Nb 3+ , Pr 4+ );
( Ge 3+ , Se 5+ );	(Kr 6+ , Y 7+ );	(Nb 3+ , Tb 4+ );
( Ge 3+ , Pb 5+ );	(Rb 1+ , Nb 3+ );	(Nb 4+ , N 4+ );
( As 1+ , Sc 2+ );	(Rb 2+ , Te 4+ );	(Mo 1+ , Ba 2+ );
( As 1+ , Y 2+ );	(Rb 2+ , Sm 4+ );	(Mo 1+ , Pr 2+ );
( As 1+ , Zr 2+ );	(Rb 2+ , Gd 4+ );	(Mo 1+ , Nd 2+ );
( As 1+ , Lu 2+ );	(Rb 2+ , Dy 4+ );	(Mo 1+ , Ra 2+ );
( As 2+ , Co 3+ );	(Rb 2+ , Ho 4+ );	(Mo 2+ , Ru 3+ );
( As 2+ , Ga 3+ );	(Rb 2+ , Er 4+ );	(Mo 2+ , Sn 3+ );
( As 2+ , Ge 3+ );	(Rb 3+ , Mg 3+ );	(Mo 3+ , Cr 4+ );
( As 2+ , Tl 3+ );	(Rb 3+ , Te 5+ );	(Mo 3+ , Ge 4+ );
( As 3+ , Fe 4+ );	(Rb 5+ , Rb 6+ );	(Mo 4+ , Bi 5+ );
( As 4+ , Sb 6+ );	(Rb 6+ , Te 7+ );	(Mo 5+ , Mn 6+ );
( Se 1+ , Al 2+ );	(Sr 1+ , Be 2+ );	(Mo 6+ , O 6+ );
( Se 1+ , Si 2+ );	(Sr 1+ , Zn 2+ );	(Mo 6+ , Cr 7+ );
( Tc 1+ , Sr 2+ );	(Sn 1+ , Er 2+ );	(Pr 2+ , Xe 2+ );
( Tc 1+ , La 2+ );	(Sn 2+ , N 2+ );	(Pr 2+ , Pr 3+ );
( Tc 1+ , Ce 2+ );	(Sn 2+ , Ar 2+ );	(Pr 2+ , Nd 3+ );
( Tc 1+ , Pm 2+ );	(Sn 2+ , V 3+ );	(Pr 2+ , Pm 3+ );
( Tc 1+ , Sm 2+ );	(Sn 2+ , Mo 3+ );	(Pr 2+ , Gd 3+ );
( Tc 1+ , Eu 2+ );	(Sn 3+ , Mn 4+ );	(Pr 2+ , Tb 3+ );
( Tc 1+ , Tb 2+ );	(Sn 3+ , Fe 4+ );	(Nd 2+ , Sm 3+ );
( Tc 1+ , Dy 2+ );	(Sn 3+ , Co 4+ );	(Nd 2+ , Dy 3+ );
( Ru 1+ , Ca 2+ );	(Sb 2+ , Ti 3+ );	(Nd 2+ , Ho 3+ );
( Ru 1+ , Eu 2+ );	(Sb 2+ , Sb 3+ );	(Nd 2+ , Er 3+ );
( Ru 1+ , Tb 2+ );	(Sb 2+ , Bi 3+ );	(Nd 2+ , Lu 3+ );
( Ru 1+ , Dy 2+ );	(Sb 3+ , C 3+ );	(Pm 2+ , C 2+ );
( Ru 1+ , Ho 2+ );	(Te 1+ , Sc 2+ );	(Pm 2+ , K 2+ );
( Ru 1+ , Er 2+ );	(Te 1+ , Y 2+ );	(Pm 2+ , Zr 3+ );
( Rh 1+ , V 2+ );	(Te 1+ , Gd 2+ );	(Pm 2+ , Eu 3+ );
( Rh 1+ , Nb 2+ );	(Te 1+ , Tm 2+ );	(Pm 2+ , Tm 3+ );
( Rh 1+ , Sn 2+ );	(Te 1+ , Yb 2+ );	(Sm 2+ , Cl 2+ );
( Rh 1+ , Hf 2+ );	(Te 1+ , Lu 2+ );	(Sm 2+ , Sc 3+ );
( Pd 1+ , Al 2+ );	(Te 2+ , Sc 3+ );	(Sm 2+ , Yb 3+ );
( Pd 1+ , Si 2+ );	(Te 2+ , Kr 2+ );	(Eu 2+ , Nb 3+ );
( Pd 1+ , Fe 2+ );	(Te 2+ , Yb 3+ );	(Gd 2+ , Cl 2+ );
( Pd 1+ , Co 2+ );	(Te 2+ , Hf 3+ );	(Gd 2+ , Sc 3+ );
( Pd 1+ , Ru 2+ );	(Te 3+ , Ar 3+ );	(Gd 2+ , Eu 3+ );
( Pd 1+ , In 2+ );	(Te 3+ , La 4+ );	(Gd 2+ , Yb 3+ );
( Pd 1+ , Sb 2+ );	(Te 3+ , Yb 4+ );	(Tb 2+ , B 2+ );
( Pd 1+ , Bi 2+ );	(Te 4+ , Bi 5+ );	(Tb 2+ , S 2+ );
( Ag 1+ , Cu 2+ );	(I 1+ , Al 2+ );	(Tb 2+ , Br 2+ );
( Ag 1+ , As 2+ );	(I 1+ , Si 2+ );	(Tb 2+ , Xe 2+ );
( Ag 1+ , Ag 2+ );	(I 1+ , Fe 2+ );	(Tb 2+ , Sm 3+ );
( Ag 1+ , Cs 2+ );	(I 1+ , Co 2+ );	(Tb 2+ , Tb 3+ );

( Ag 1+ , Hg 2+ );	(I 1+ , Ge 2+ );	(Tb 2+ , Dy 3+ );
( Cd 1+ , Zn 2+ );	(I 1+ , Ru 2+ );	(Tb 2+ , Ho 3+ );
( Cd 1+ , Ga 2+ );	(I 1+ , In 2+ );	(Tb 2+ , Er 3+ );
( Cd 1+ , Cd 2+ );	(I 1+ , Bi 2+ );	(Dy 2+ , Cl 2+ );
( Cd 1+ , Tl 2+ );	(Xe 1+ , Al 2+ );	(Dy 2+ , K 2+ );
( In 1+ , Sc 2+ );	(Xe 1+ , Co 2+ );	(Dy 2+ , Zr 3+ );
( In 1+ , Y 2+ );	(Xe 1+ , Ni 2+ );	(Dy 2+ , Eu 3+ );
( In 1+ , Yb 2+ );	(Xe 1+ , Rh 2+ );	(Dy 2+ , Yb 3+ );
( In 1+ , Lu 2+ );	(Xe 1+ , Cd 2+ );	(Ho 2+ , Sc 3+ );
( In 2+ , Sr 3+ );	(Xe 1+ , Sb 2+ );	(Ho 2+ , Yb 3+ );
( In 2+ , Cd 3+ );	(La 2+ , Ti 3+ );	(Ho 2+ , Hf 3+ );
( Sn 1+ , Ca 2+ );	(La 2+ , Sb 3+ );	(Er 2+ , Sc 3+ );
( Sn 1+ , Sr 2+ );	(Ce 2+ , Ag 2+ );	(Er 2+ , Yb 3+ );
( Sn 1+ , La 2+ );	(Ce 2+ , I 2+ );	(Er 2+ , Hf 3+ );
( Sn 1+ , Sm 2+ );	(Ce 2+ , Cs 2+ );	(Tm 2+ , Kr 2+ );
( Sn 1+ , Eu 2+ );	(Ce 2+ , Au 2+ );	(Tm 2+ , Nb 3+ );
( Sn 1+ , Tb 2+ );	(Ce 2+ , Hg 2+ );	(Tm 2+ , Hf 3+ );
( Sn 1+ , Dy 2+ );	(Pr 2+ , B 2+ );	(Yb 2+ , Ti 3+ );
( Sn 1+ , Ho 2+ );	(Pr 2+ , Y 3+ );	(Lu 2+ , Kr 2+ );
( Lu 2+ , Hf 3+ );	(Pb 2+ , As 3+ );	(Tl 1+ , Mg 2+ );
( Hf 2+ , As 2+ );	(Pb 2+ , In 3+ );	(Tl 1+ , Mn 2+ );
( Hf 2+ , Ag 2+ );	(Pb 2+ , Te 3+ );	(Tl 1+ , Mo 2+ );
( Hf 2+ , I 2+ );	(Pb 2+ , Pb 3+ );	(Tl 1+ , Tc 2+ );
( Hf 2+ , Cs 2+ );	(Pb 3+ , Br 4+ );	(Tl 1+ , Sn 2+ );
( Hf 2+ , Hg 2+ );	(Bi 1+ , Ba 2+ );	(Tl 1+ , Pb 2+ );
( Hg 1+ , Al 2+ );	(Bi 2+ , Ar 2+ );	(Pb 1+ , Sc 2+ );
( Hg 1+ , Si 2+ );	(Bi 2+ , Mo 3+ );	(Pb 1+ , Y 2+ );
( Hg 1+ , Co 2+ );	(Bi 3+ , Se 4+ );	(Pb 1+ , Lu 2+ );
( Hg 1+ , Ni 2+ );	(Bi 3+ , Mo 4+ );	(Pb 2+ , Fe 3+ );
( Hg 1+ , Rh 2+ );	(Bi 3+ , Pb 4+ );	
( Hg 1+ , Cd 2+ );	(Bi 4+ , P 5+ );	
( Hg 1+ , In 2+ );	(Bi 4+ , Kr 5+ );	
( Hg 1+ , Sb 2+ );	(Bi 4+ , Zr 5+ );	

and

209. (Previously Presented) A method according to claim 193, further comprising forming said hydrido atoms from hydrogen atoms by use of a catalyst comprising at least one free atom selected from the group consisting of Be, Cu, Zn, Pd, Te and Pt.
210. (Previously Presented) A method according to claim 193, further comprising forming hydrido atoms from hydrogen atoms by use of a catalyst comprising at least one set of two species selected from the group consisting of:

( Li 0+ , Ar 5+ ); ( P 1+ , Nd 4+ ); ( Ti 2+ , As 5+ );  
 ( Li 0+ , Mo 6+ ); ( P 1+ , Tb 4+ ); ( Ti 2+ , Se 5+ );  
 ( Be 0+ , Kr 5+ ); ( P 3+ , Na 5+ ); ( V 1+ , Cd 3+ );  
 ( B 0+ , Sc 3+ ); ( S 0+ , Sm 3+ ); ( V 1+ , I 3+ );  
 ( B 0+ , Zr 3+ ); ( S 0+ , Dy 3+ ); ( V 1+ , Hg 3+ );  
 ( B 0+ , Yb 3+ ); ( S 0+ , Ho 3+ ); ( V 2+ , Kr 4+ );  
 ( C 0+ , Te 3+ ); ( S 0+ , Er 3+ ); ( V 2+ , Nb 5+ );  
 ( C 0+ , Tl 3+ ); ( S 0+ , Lu 3+ ); ( V 4+ , Ni 7+ );  
 ( N 0+ , Ag 3+ ); ( S 1+ , Nb 4+ ); ( V 4+ , Kr 8+ );  
 ( N 0+ , Cd 3+ ); ( S 1+ , Ho 4+ ); ( Cr 1+ , S 3+ );  
 ( N 0+ , Hg 3+ ); ( S 1+ , Er 4+ ); ( Cr 1+ , Ca 3+ );  
 ( N 1+ , Bi 5+ ); ( S 1+ , Tm 4+ ); ( Cr 3+ , Be 3+ );  
 ( N 2+ , Br 6+ ); ( S 2+ , Bi 5+ ); ( Cr 3+ , Zn 5+ );  
 ( N 2+ , Kr 6+ ); ( Cl 0+ , Ti 3+ ); ( Cr 5+ , Cu 8+ );  
 ( O 0+ , Cl 3+ ); ( Cl 1+ , Mo 4+ ); ( Mn 1+ , Nd 4+ );  
 ( O 0+ , Kr 3+ ); ( Cl 1+ , Pb 4+ ); ( Mn 1+ , Tb 4+ );  
 ( O 0+ , Sm 4+ ); ( Cl 3+ , Sc 5+ ); ( Mn 2+ , Ca 4+ );  
 ( O 0+ , Dy 4+ ); ( Cl 4+ , Br 7+ ); ( Mn 3+ , Nb 6+ );  
 ( O 2+ , Na 4+ ); ( Ar 0+ , Mn 3+ ); ( Mn 5+ , Ca 8+ );  
 ( O 2+ , Cl 6+ ); ( Ar 0+ , As 3+ ); ( Fe 1+ , Nd 4+ );  
 ( O 2+ , Mn 6+ ); ( Ar 0+ , Rh 3+ ); ( Fe 1+ , Pm 4+ );  
 ( O 3+ , Al 5+ ); ( Ar 0+ , Tl 3+ ); ( Fe 1+ , Tb 4+ );  
 ( F 0+ , Bi 4+ ); ( Ar 1+ , Mn 4+ ); ( Fe 3+ , Ne 4+ );  
 ( F 1+ , Mn 5+ ); ( Ar 1+ , In 4+ ); ( Fe 5+ , Mo 8+ );  
 ( F 3+ , Mg 5+ ); ( Ar 5+ , Mg 5+ ); ( Co 1+ , Pm 4+ );  
 ( F 4+ , Ti 8+ ); ( K 0+ , Al 3+ ); ( Co 2+ , C 4+ );  
 ( Ne 1+ , Ge 5+ ); ( K 0+ , Cr 3+ ); ( Co 3+ , Mg 4+ );  
 ( Ne 4+ , Al 6+ ); ( K 0+ , Pb 3+ ); ( Ni 1+ , La 4+ );  
 ( Na 0+ , Cr 4+ ); ( K 1+ , Sc 4+ ); ( Ni 1+ , Yb 4+ );  
 ( Na 0+ , Ge 4+ ); ( K 2+ , Cl 5+ ); ( Ni 1+ , Lu 4+ );  
 ( Na 1+ , Sc 5+ ); ( Ca 0+ , Eu 3+ ); ( Ni 2+ , K 4+ );  
 ( Na 1+ , Bi 6+ ); ( Ca 0+ , Dy 3+ ); ( Ni 5+ , Fe 8+ );  
 ( Na 3+ , Ne 6+ ); ( Ca 0+ , Ho 3+ ); ( Cu 0+ , Ce 3+ );  
 ( Na 4+ , Ne 7+ ); ( Ca 0+ , Er 3+ ); ( Cu 0+ , Pr 3+ );  
 ( Mg 0+ , Kr 3+ ); ( Ca 1+ , Mg 3+ ); ( Cu 1+ , Ar 3+ );  
 ( Mg 2+ , Al 5+ ); ( Ca 1+ , Fe 4+ ); ( Cu 1+ , Ti 4+ );  
 ( Mg 3+ , Na 6+ ); ( Ca 1+ , Co 4+ ); ( Cu 1+ , Te 4+ );  
 ( Al 1+ , Zr 5+ ); ( Ca 3+ , Co 6+ ); ( Cu 2+ , Sn 5+ );  
 ( Al 3+ , Mg 6+ ); ( Ca 3+ , Y 6+ ); ( Zn 0+ , Y 3+ );  
 ( Al 3+ , Cr 8+ ); ( Sc 1+ , C 3+ ); ( Zn 0+ , Pm 3+ );  
 ( Si 1+ , Zn 3+ ); ( Sc 1+ , Te 4+ ); ( Zn 0+ , Gd 3+ );  
 ( Si 1+ , Ce 4+ ); ( Ti 1+ , Mn 3+ ); ( Zn 0+ , Tb 3+ );  
 ( Si 2+ , Na 4+ ); ( Ti 1+ , Ga 3+ ); ( Zn 1+ , Mo 4+ );  
 ( Si 2+ , Cl 6+ ); ( Ti 1+ , As 3+ ); ( Zn 1+ , Pb 4+ );  
 ( Si 3+ , Be 4+ ); ( Ti 1+ , Rh 3+ ); ( Zn 2+ , N 4+ );  
 ( Si 5+ , N 6+ ); ( Ti 1+ , Tl 3+ ); ( Zn 2+ , Kr 5+ );  
 ( Zn 3+ , N 5+ ); ( Y 5+ , Co 7+ ); ( Ce 1+ , Ho 3+ );  
 ( Zn 5+ , Co 8+ ); ( Zr 1+ , Zr 3+ ); ( Ce 1+ , Er 3+ );  
 ( Ga 1+ , Bi 4+ ); ( Zr 2+ , Sc 4+ ); ( Ce 1+ , Lu 3+ );



( Ge 1+ , S 3+ ); ( Zr 2+ , Sr 4+ ); ( Pr 1+ , Sc 3+ );  
 ( Ge 1+ , Ce 4+ ); ( Nb 1+ , Mo 3+ ); ( Pr 1+ , Zr 3+ );  
 ( As 1+ , Ca 3+ ); ( Nb 1+ , Sb 3+ ); ( Pr 1+ , Yb 3+ );  
 ( As 1+ , Br 3+ ); ( Nb 1+ , Bi 3+ ); ( Nd 1+ , Nb 3+ );  
 ( As 2+ , F 3+ ); ( Nb 2+ , Sn 4+ ); ( Nd 1+ , Hf 3+ );  
 ( As 2+ , Kr 4+ ); ( Nb 2+ , Sb 4+ ); ( Pm 1+ , Nb 3+ );  
 ( As 2+ , Nb 5+ ); ( Nb 3+ , Co 5+ ); ( Sm 1+ , Ti 3+ );  
 ( Se 1+ , Zn 3+ ); ( Nb 3+ , Rb 5+ ); ( Eu 1+ , V 3+ );  
 ( Se 1+ , Ce 4+ ); ( Nb 4+ , Zn 6+ ); ( Eu 1+ , Mo 3+ );  
 ( Se 2+ , Kr 4+ ); ( Mo 1+ , Se 3+ ); ( Eu 1+ , Sb 3+ );  
 ( Se 2+ , Nb 5+ ); ( Mo 1+ , I 3+ ); ( Gd 1+ , Bi 3+ );  
 ( Se 3+ , Ni 5+ ); ( Mo 4+ , Fe 6+ ); ( Tb 1+ , Hf 3+ );  
 ( Se 4+ , Nb 7+ ); ( Mo 5+ , Rb 8+ ); ( Dy 1+ , Ti 3+ );  
 ( Br 0+ , Eu 3+ ); ( Ag 0+ , La 3+ ); ( Ho 1+ , Bi 3+ );  
 ( Br 0+ , Tm 3+ ); ( Ag 0+ , Ce 3+ ); ( Er 1+ , Bi 3+ );  
 ( Br 1+ , Nb 4+ ); ( Cd 0+ , La 3+ ); ( Tm 1+ , V 3+ );  
 ( Br 1+ , Gd 4+ ); ( In 1+ , Nd 4+ ); ( Tm 1+ , Mo 3+ );  
 ( Br 1+ , Ho 4+ ); ( In 1+ , Tb 4+ ); ( Tm 1+ , Sb 3+ );  
 ( Br 1+ , Er 4+ ); ( Sn 1+ , Si 3+ ); ( Yb 1+ , Al 3+ );  
 ( Br 2+ , F 3+ ); ( Sn 1+ , Co 3+ ); ( Yb 1+ , Ru 3+ );  
 ( Br 2+ , Ga 4+ ); ( Sn 1+ , Ge 3+ ); ( Yb 1+ , In 3+ );  
 ( Br 3+ , O 4+ ); ( Sn 2+ , F 3+ ); ( Yb 1+ , Sn 3+ );  
 ( Br 3+ , Al 4+ ); ( Sn 2+ , Ga 4+ ); ( Lu 1+ , Tc 3+ );  
 ( Br 4+ , N 5+ ); ( Sb 1+ , Si 3+ ); ( Lu 1+ , Ru 3+ );  
 ( Kr 0+ , Ti 3+ ); ( Sb 1+ , Co 3+ ); ( Lu 1+ , In 3+ );  
 ( Kr 1+ , Sn 4+ ); ( Sb 1+ , Ge 3+ ); ( Lu 1+ , Sn 3+ );  
 ( Kr 1+ , Sb 4+ ); ( Sb 2+ , As 4+ ); ( Hf 1+ , Sc 3+ );  
 ( Kr 2+ , Ne 3+ ); ( Te 1+ , Mn 3+ ); ( Hf 1+ , Yb 3+ );  
 ( Kr 2+ , Bi 5+ ); ( Te 1+ , As 3+ ); ( Hg 0+ , La 3+ );  
 ( Kr 3+ , O 4+ ); ( Te 1+ , Rh 3+ ); ( Pb 1+ , Ni 3+ );  
 ( Kr 3+ , Al 4+ ); ( Te 1+ , Te 3+ ); ( Pb 1+ , Se 3+ );  
 ( Kr 4+ , Ar 6+ ); ( Te 1+ , Tl 3+ ); ( Pb 2+ , F 3+ );  
 ( Rb 0+ , Sc 3+ ); ( Te 2+ , Cr 4+ ); ( Pb 2+ , Ga 4+ );  
 ( Rb 0+ , Zr 3+ ); ( Te 2+ , Ge 4+ ); ( Bi 1+ , P 3+ );  
 ( Rb 0+ , Yb 3+ ); ( Te 2+ , As 4+ ); ( Bi 1+ , Sr 3+ );  
 ( Rb 1+ , N 3+ ); ( Te 3+ , Zr 5+ ); ( La 1+ , Ru 3+ );  
 ( Sr 1+ , C 3+ ); ( Te 4+ , Ni 6+ ); ( La 1+ , In 3+ );  
 ( Sr 1+ , Ar 3+ ); ( Te 4+ , Cu 6+ ); ( La 1+ , Sn 3+ );  
 ( Sr 1+ , Ti 4+ ); ( Xe 0+ , Pr 3+ ); ( Ce 1+ , Sm 3+ ); and  
 ( Sr 1+ , Te 4+ ); ( Xe 0+ , Nd 3+ ); ( Ce 1+ , Dy 3+ );  
 ( Sr 3+ , Nb 6+ ); ( La 1+ , Tc 3+ );

211. (Previously Presented) A method according to claim 193, further comprising forming said hydrino atoms from hydrogen atoms by use of a catalyst comprising at least one pair selected from the group consisting of:

( Ne 1+ , H 1+ ),	( Kr 3+ , B 2+ ),	( Tm 3+ , N 1+ ),
( Ar 2+ , H 1+ ),	( Rb 3+ , B 2+ ),	( Pb 3+ , N 1+ ),
( Sn 3+ , H 1+ ),	( B 2+ , P 1+ ),	( Sr 3+ , N 2+ ),
( Pm 3+ , H 1+ ),	( P 4+ , B 3+ ),	( N 2+ , P 2+ ),
( Sm 3+ , H 1+ ),	( B 2+ , S 1+ ),	( Ar 4+ , N 3+ ),
( Dy 3+ , H 1+ ),	( V 4+ , B 3+ ),	( Fe 4+ , N 3+ ),
( Kr 3+ , He 1+ ),	( B 2+ , As 1+ ),	( Ni 4+ , N 3+ ),
( Rb 3+ , He 1+ ),	( B 2+ , Se 1+ ),	( N 2+ , Cu 2+ ),
( K 4+ , He 2+ ),	( B 2+ , I 1+ ),	( N 2+ , Pd 2+ ),
( Zn 4+ , He 2+ ),	( B 2+ , Ba 2+ ),	( N 2+ , I 2+ ),
( Se 5+ , He 2+ ),	( B 2+ , Ce 2+ ),	( N 2+ , La 3+ ),
( He 1+ , Rb 2+ ),	( B 2+ , Pr 2+ ),	( N 2+ , Ce 3+ ),
( Zr 4+ , He 2+ ),	( B 2+ , Nd 2+ ),	( N 2+ , Tl 2+ ),
( He 1+ , Mo 3+ ),	( B 2+ , Pm 2+ ),	( N 3+ , Cr 4+ ),
( Si 2+ , Li 1+ ),	( B 2+ , Hg 1+ ),	( N 3+ , As 4+ ),
( Mn 2+ , Li 1+ ),	( B 2+ , Rn 1+ ),	( N 3+ , La 4+ ),
( Co 2+ , Li 1+ ),	( B 2+ , Ra 2+ ),	( Ne 4+ , N 5+ ),
( Pd 2+ , Li 1+ ),	( Cl 2+ , C 1+ ),	( Fe 6+ , N 5+ ),
( I 2+ , Li 1+ ),	( Zn 2+ , C 1+ ),	( Kr 7+ , N 5+ ),
( Hf 3+ , Li 1+ ),	( Nb 3+ , C 1+ ),	( Nb 6+ , N 5+ ),
( Li 1+ , C 3+ ),	( Pr 3+ , C 1+ ),	( N 4+ , Te 6+ ),
( Li 1+ , N 3+ ),	( Kr 3+ , C 2+ ),	( Ne 1+ , O 1+ ),
( Li 1+ , Na 2+ ),	( Rb 3+ , C 2+ ),	( Ar 2+ , O 1+ ),
( Li 1+ , S 4+ ),	( C 2+ , P 2+ ),	( Sn 3+ , O 1+ ),
( Cu 5+ , Li 2+ ),	( Ar 4+ , C 3+ ),	( Pm 3+ , O 1+ ),
( Li 1+ , Br 4+ ),	( Fe 4+ , C 3+ ),	( Sm 3+ , O 1+ ),
( Br 6+ , Li 2+ ),	( Ni 4+ , C 3+ ),	( Dy 3+ , O 1+ ),
( V 6+ , Li 3+ ),	( C 2+ , Cu 2+ ),	( F 2+ , O 2+ ),
( Li 2+ , Mn 6+ ),	( C 2+ , Ga 2+ ),	( Ne 2+ , O 2+ ),
( Cu 2+ , Be 1+ ),	( C 2+ , Y 3+ ),	( O 1+ , Mg 1+ ),
( Kr 2+ , Be 1+ ),	( C 2+ , Pd 2+ ),	( O 1+ , Ti 1+ ),
( Cd 2+ , Be 1+ ),	( C 2+ , Ce 3+ ),	( O 1+ , V 1+ ),
( Te 3+ , Be 1+ ),	( C 2+ , Gd 3+ ),	( O 1+ , Cr 1+ ),
( Ce 3+ , Be 1+ ),	( C 2+ , Au 2+ ),	( O 1+ , Mn 1+ ),
( K 2+ , Be 2+ ),	( C 2+ , Tl 2+ ),	( O 1+ , Fe 1+ ),
( V 3+ , Be 2+ ),	( Sc 4+ , C 4+ ),	( O 1+ , Co 1+ ),
( Ge 3+ , Be 2+ ),	( C 3+ , Cu 3+ ),	( O 1+ , Ni 1+ ),
( Mo 3+ , Be 2+ ),	( C 3+ , Br 3+ ),	( O 1+ , Cu 1+ ),
( Bi 3+ , Be 2+ ),	( C 3+ , Kr 3+ ),	( O 1+ , Ge 1+ ),
( Be 2+ , Ne 5+ ),	( C 3+ , Cd 3+ ),	( O 1+ , Zr 1+ ),
( Be 2+ , Kr 8+ ),	( C 3+ , Te 4+ ),	( O 1+ , Nb 1+ ),
( Be 2+ , Mo 7+ ),	( C 3+ , Ce 4+ ),	( O 1+ , Mo 1+ ),
( Be 3+ , Al 6+ ),	( Se 3+ , N 1+ ),	( O 1+ , Tc 1+ ),
( Br 2+ , B 1+ ),	( Eu 3+ , N 1+ ),	( O 1+ , Ru 1+ ),
( Ce 3+ , B 1+ ),	( Ho 3+ , N 1+ ),	( O 1+ , Rh 1+ ),
( Cl 3+ , B 2+ ),	( Er 3+ , N 1+ ),	( O 1+ , Ag 1+ ),
( O 1+ , Sn 1+ ),	( Ar 5+ , F 3+ ),	( Hf 3+ , Na 1+ ),
( O 1+ , Ta 1+ ),	( Cr 5+ , F 3+ ),	( Na 1+ , Al 2+ ),

( O 1+ , W 1+ ),	( F 2+ , Ni 3+ ),	( Na 1+ , P 2+ ),
( O 1+ , Re 1+ ),	( F 2+ , Ge 3+ ),	( Ar 4+ , Na 2+ ),
( O 1+ , Pb 1+ ),	( Sr 5+ , F 3+ ),	( Fe 4+ , Na 2+ ),
( O 1+ , Bi 1+ ),	( F 2+ , Zr 4+ ),	( Ni 4+ , Na 2+ ),
( O 2+ , Ar 2+ ),	( F 2+ , Ag 3+ ),	( Na 1+ , Pd 2+ ),
( K 4+ , O 3+ ),	( F 4+ , F 4+ ),	( Na 1+ , In 2+ ),
( O 2+ , Ti 3+ ),	( Cl 6+ , F 4+ ),	( Na 1+ , I 2+ ),
( Zn 4+ , O 3+ ),	( F 3+ , Ar 4+ ),	( Na 1+ , La 3+ ),
( O 2+ , Rb 2+ ),	( F 3+ , Zn 4+ ),	( Na 1+ , Ce 3+ ),
( O 2+ , Mo 3+ ),	( F 3+ , Br 5+ ),	( Na 3+ , Na 3+ ),
( O 3+ , Cr 4+ ),	( F 3+ , Te 5+ ),	( K 5+ , Na 3+ ),
( O 3+ , As 4+ ),	( F 4+ , F 4+ ),	( Na 2+ , Ti 4+ ),
( O 3+ , La 4+ ),	( Mg 4+ , F 5+ ),	( Ti 4+ , Na 3+ ),
( Mg 4+ , O 5+ ),	( F 6+ , F 6+ ),	( Fe 5+ , Na 3+ ),
( O 5+ , Sc 6+ ),	( Cr 7+ , F 6+ ),	( Rb 6+ , Na 3+ ),
( Cu 7+ , O 6+ ),	( F 5+ , Co 7+ ),	( Na 2+ , Sr 3+ ),
( O 5+ , Kr 7+ ),	( F 5+ , Y 8+ ),	( Na 2+ , Sb 4+ ),
( Si 3+ , F 1+ ),	( F 6+ , F 6+ ),	( Na 2+ , Gd 4+ ),
( K 2+ , F 1+ ),	( F 6+ , Ne 6+ ),	( Na 2+ , Yb 4+ ),
( Ge 3+ , F 1+ ),	( F 6+ , Co 8+ ),	( Na 3+ , Na 3+ ),
( Lu 3+ , F 1+ ),	( Cr 3+ , Ne 1+ ),	( Kr 7+ , Na 4+ ),
( Bi 3+ , F 1+ ),	( La 3+ , Ne 1+ ),	( Na 3+ , Rb 5+ ),
( F 2+ , F 2+ ),	( Ne 1+ , Cl 1+ ),	( Na 3+ , Sr 5+ ),
( Ne 2+ , F 2+ ),	( Ne 1+ , Sc 2+ ),	( Mo 6+ , Na 4+ ),
( F 1+ , Mg 1+ ),	( Ne 1+ , Ti 2+ ),	( Na 3+ , Te 6+ ),
( F 1+ , Sc 1+ ),	( Cr 4+ , Ne 2+ ),	( Si 4+ , Na 5+ ),
( F 1+ , Ti 1+ ),	( Se 4+ , Ne 2+ ),	( Na 4+ , Sc 6+ ),
( F 1+ , V 1+ ),	( Ne 1+ , Zr 2+ ),	( Cu 7+ , Na 5+ ),
( F 1+ , Cr 1+ ),	( Mo 5+ , Ne 2+ ),	( Na 4+ , Kr 7+ ),
( F 1+ , Mn 1+ ),	( Ne 1+ , Lu 2+ ),	( S 2+ , Mg 1+ ),
( F 1+ , Fe 1+ ),	( Pb 4+ , Ne 2+ ),	( Ni 2+ , Mg 1+ ),
( F 1+ , Co 1+ ),	( Ar 5+ , Ne 3+ ),	( Br 2+ , Mg 1+ ),
( F 1+ , Ni 1+ ),	( Sc 4+ , Ne 3+ ),	( Ag 2+ , Mg 1+ ),
( F 1+ , Cu 1+ ),	( Cr 5+ , Ne 3+ ),	( Ti 3+ , Mg 2+ ),
( F 1+ , Ge 1+ ),	( Ne 2+ , Ni 3+ ),	( Se 3+ , Mg 2+ ),
( F 1+ , Zr 1+ ),	( Ne 2+ , Br 3+ ),	( Eu 3+ , Mg 2+ ),
( F 1+ , Nb 1+ ),	( Sr 5+ , Ne 3+ ),	( Ho 3+ , Mg 2+ ),
( F 1+ , Mo 1+ ),	( Ar 6+ , Ne 4+ ),	( Er 3+ , Mg 2+ ),
( F 1+ , Tc 1+ ),	( Ne 3+ , Cr 5+ ),	( Tm 3+ , Mg 2+ ),
( F 1+ , Ru 1+ ),	( Fe 6+ , Ne 4+ ),	( Pb 3+ , Mg 2+ ),
( F 1+ , Rh 1+ ),	( Nb 6+ , Ne 4+ ),	( Ni 5+ , Mg 3+ ),
( F 1+ , Ag 1+ ),	( Ne 3+ , Pb 5+ ),	( Zn 5+ , Mg 3+ ),
( F 1+ , Sn 1+ ),	( Ne 4+ , Na 4+ ),	( Mg 2+ , Kr 4+ ),
( F 1+ , Hf 1+ ),	( Al 4+ , Ne 5+ ),	( Mg 2+ , Rb 4+ ),
( F 1+ , Ta 1+ ),	( Ne 4+ , Fe 6+ ),	( Sb 5+ , Mg 3+ ),
( F 1+ , Re 1+ ),	( Ne 4+ , Rb 7+ ),	( Mg 3+ , Se 6+ ),
( F 1+ , Pb 1+ ),	( Si 2+ , Na 1+ ),	( Mg 3+ , Zr 5+ ),
( F 1+ , Bi 1+ ),	( Co 2+ , Na 1+ ),	( Te 6+ , Mg 4+ ),
( F 2+ , F 2+ ),	( Pd 2+ , Na 1+ ),	( Mg 4+ , Cl 7+ ),

( F 2 + , S 3 + ),	( I 2 + , Na 1 + ),	( Ti 7 + , Mg 5 + ),
( Mg 5 + , Sc 8 + ),	( Si 2 + , Ac 1 + ),	( S 2 + , Ti 1 + ),
( Mg 6 + , Mn 8 + ),	( Si 2 + , Th 1 + ),	( S 2 + , V 1 + ),
( Si 2 + , Al 1 + ),	( Si 2 + , Pa 1 + ),	( S 2 + , Cr 1 + ),
( Mn 2 + , Al 1 + ),	( Si 2 + , U 1 + ),	( S 2 + , Mn 1 + ),
( Co 2 + , Al 1 + ),	( Si 2 + , Np 1 + ),	( S 2 + , Ni 1 + ),
( Ge 2 + , Al 1 + ),	( Si 2 + , Pu 1 + ),	( S 2 + , Cu 1 + ),
( Zr 3 + , Al 1 + ),	( Si 2 + , Am 1 + ),	( S 2 + , Y 1 + ),
( I 2 + , Al 1 + ),	( Si 2 + , Cm 1 + ),	( S 2 + , Zr 1 + ),
( Hf 3 + , Al 1 + ),	( Si 2 + , Bk 1 + ),	( S 2 + , Nb 1 + ),
( Hg 2 + , Al 1 + ),	( Si 2 + , Cf 1 + ),	( S 2 + , Mo 1 + ),
( S 3 + , Al 2 + ),	( Si 2 + , Es 1 + ),	( S 2 + , Tc 1 + ),
( V 3 + , Al 2 + ),	( S 4 + , Si 4 + ),	( S 2 + , Ru 1 + ),
( Br 3 + , Al 2 + ),	( Sc 3 + , Si 4 + ),	( S 2 + , Rh 1 + ),
( Mo 3 + , Al 2 + ),	( Mn 4 + , Si 4 + ),	( S 2 + , Ag 1 + ),
( Sb 4 + , Al 3 + ),	( Si 3 + , Co 2 + ),	( S 2 + , Sn 1 + ),
( Bi 4 + , Al 3 + ),	( Si 3 + , Zn 2 + ),	( S 2 + , Hf 1 + ),
( Ca 7 + , Al 4 + ),	( Si 3 + , Ru 2 + ),	( S 2 + , Pb 1 + ),
( Al 3 + , Sc 5 + ),	( Si 3 + , Rh 2 + ),	( S 2 + , Bi 1 + ),
( Al 4 + , Kr 8 + ),	( Si 3 + , Cd 2 + ),	( S 2 + , Es 1 + ),
( Al 5 + , Ni 8 + ),	( Sn 4 + , Si 4 + ),	( Ar 4 + , S 4 + ),
( Ni 2 + , Si 1 + ),	( Si 3 + , Bi 2 + ),	( Fe 4 + , S 4 + ),
( Br 2 + , Si 1 + ),	( Si 4 + , Cu 7 + ),	( Ni 4 + , S 4 + ),
( Sr 2 + , Si 2 + ),	( Nb 3 + , P 1 + ),	( S 3 + , Cu 2 + ),
( Sb 3 + , Si 2 + ),	( Pr 3 + , P 1 + ),	( S 3 + , Pd 2 + ),
( Gd 3 + , Si 2 + ),	( S 3 + , P 2 + ),	( S 3 + , In 2 + ),
( Yb 3 + , Si 2 + ),	( Br 3 + , P 2 + ),	( S 3 + , I 2 + ),
( K 3 + , Si 3 + ),	( P 3 + , S 2 + ),	( S 3 + , La 3 + ),
( Si 2 + , Ca 1 + ),	( P 3 + , Cl 2 + ),	( S 3 + , Ce 3 + ),
( Si 2 + , Ga 1 + ),	( Co 4 + , P 4 + ),	( K 5 + , S 5 + ),
( Si 2 + , Sr 1 + ),	( P 3 + , Kr 2 + ),	( S 4 + , Sb 4 + ),
( Si 2 + , Y 1 + ),	( Kr 5 + , P 4 + ),	( S 4 + , Lu 4 + ),
( Y 3 + , Si 3 + ),	( P 3 + , Zr 3 + ),	( S 4 + , Bi 4 + ),
( Mo 4 + , Si 3 + ),	( P 3 + , Sm 3 + ),	( S 5 + , Ar 4 + ),
( Si 2 + , In 1 + ),	( P 3 + , Tm 3 + ),	( S 5 + , K 4 + ),
( Si 2 + , Ba 1 + ),	( P 3 + , Hf 3 + ),	( S 5 + , Br 5 + ),
( Si 2 + , La 1 + ),	( P 4 + , Cu 3 + ),	( Y 6 + , S 6 + ),
( Si 2 + , Ce 1 + ),	( Ge 4 + , P 5 + ),	( Ar 2 + , Cl 1 + ),
( Si 2 + , Pr 1 + ),	( P 4 + , Kr 3 + ),	( Rb 2 + , Cl 1 + ),
( Si 2 + , Nd 1 + ),	( Y 5 + , P 5 + ),	( Sn 3 + , Cl 1 + ),
( Si 2 + , Pm 1 + ),	( P 4 + , Cd 3 + ),	( Nd 3 + , Cl 1 + ),
( Si 2 + , Sm 1 + ),	( P 4 + , Te 4 + ),	( Pm 3 + , Cl 1 + ),
( Si 2 + , Eu 1 + ),	( P 4 + , Ce 4 + ),	( Sm 3 + , Cl 1 + ),
( Si 2 + , Gd 1 + ),	( P 5 + , Br 8 + ),	( Ca 2 + , Cl 2 + ),
( Si 2 + , Tb 1 + ),	( P 7 + , S 7 + ),	( Mn 3 + , Cl 2 + ),
( Si 2 + , Dy 1 + ),	( Nb 3 + , S 1 + ),	( Co 3 + , Cl 2 + ),
( Si 2 + , Ho 1 + ),	( Cd 2 + , S 1 + ),	( Cl 4 + , Cl 3 + ),
( Si 2 + , Er 1 + ),	( Te 3 + , S 1 + ),	( Cl 2 + , Ca 2 + ),
( Si 2 + , Tm 1 + ),	( Ca 2 + , S 2 + ),	( Ca 3 + , Cl 3 + ),

( Si 2 + , Yb 1 + ),	( Mn 3 + , S 2 + ),	( Cl 2 + , Br 1 + ),
( Si 2 + , Lu 1 + ),	( Co 3 + , S 2 + ),	( Cl 2 + , Y 2 + ),
( Si 2 + , Tl 1 + ),	( Nb 4 + , S 2 + ),	( Mo 5 + , Cl 3 + ),
( Si 2 + , Ra 1 + ),	( S 2 + , Sc 1 + ),	( Cl 2 + , Xe 1 + ),
( Cl 2 + , Eu 2 + ),	( Br 6 + , Ar 5 + ),	( Pr 3 + , Ca 2 + ),
( Cl 2 + , Gd 2 + ),	( Nb 5 + , Ar 5 + ),	( Tb 3 + , Ca 2 + ),
( Cl 2 + , Tb 2 + ),	( Ti 5 + , Ar 6 + ),	( Kr 5 + , Ca 3 + ),
( Cl 2 + , Dy 2 + ),	( Mn 6 + , Ar 6 + ),	( Ca 2 + , Zr 3 + ),
( Cl 2 + , Ho 2 + ),	( Ar 5 + , Ga 4 + ),	( Ca 2 + , Sm 3 + ),
( Cl 2 + , Er 2 + ),	( Ar 5 + , As 5 + ),	( Ca 2 + , Dy 3 + ),
( Cl 2 + , Tm 2 + ),	( Ar 7 + , Y 7 + ),	( Ca 2 + , Ho 3 + ),
( Cl 2 + , Yb 2 + ),	( K 1 + , K 1 + ),	( Ca 2 + , Er 3 + ),
( Se 5 + , Cl 4 + ),	( Xe 2 + , K 1 + ),	( Ca 2 + , Tm 3 + ),
( Zr 4 + , Cl 4 + ),	( Pb 2 + , K 1 + ),	( Ca 2 + , Hf 3 + ),
( Cl 3 + , Nb 3 + ),	( K 1 + , K 1 + ),	( Mn 5 + , Ca 4 + ),
( Cl 3 + , Sb 3 + ),	( Zn 3 + , K 2 + ),	( Ca 3 + , Zn 3 + ),
( Cl 3 + , Cs 2 + ),	( Br 4 + , K 2 + ),	( Ca 3 + , Rb 3 + ),
( Cl 3 + , Yb 3 + ),	( K 1 + , Rb 1 + ),	( Ca 3 + , Pr 4 + ),
( Cl 3 + , Bi 3 + ),	( Te 4 + , K 2 + ),	( Ca 3 + , Tb 4 + ),
( Cl 4 + , Cl 3 + ),	( K 1 + , Cs 1 + ),	( Ca 4 + , Sr 4 + ),
( Cl 4 + , Ar 3 + ),	( Sc 3 + , K 3 + ),	( Ca 4 + , Sb 5 + ),
( Mn 5 + , Cl 5 + ),	( K 2 + , Ni 2 + ),	( Ca 4 + , Bi 5 + ),
( Cl 4 + , Zn 3 + ),	( K 2 + , Zn 2 + ),	( Ca 5 + , Se 6 + ),
( Cl 4 + , Rb 3 + ),	( K 2 + , As 2 + ),	( Rb 7 + , Ca 6 + ),
( Cl 4 + , Sn 4 + ),	( K 2 + , Rh 2 + ),	( Ca 5 + , Zr 5 + ),
( Cl 4 + , Nd 4 + ),	( K 2 + , Te 2 + ),	( Te 6 + , Ca 6 + ),
( Cl 4 + , Tb 4 + ),	( K 2 + , Pt 2 + ),	( Ca 6 + , Ti 5 + ),
( Ar 6 + , Cl 6 + ),	( K 3 + , Mn 3 + ),	( Se 6 + , Ca 7 + ),
( Cl 5 + , Cr 5 + ),	( K 3 + , Co 3 + ),	( Ca 7 + , Ti 6 + ),
( Fe 6 + , Cl 6 + ),	( Br 5 + , K 4 + ),	( Ca 7 + , Mn 7 + ),
( Nb 6 + , Cl 6 + ),	( K 3 + , Pd 3 + ),	( Mn 2 + , Sc 1 + ),
( Cl 5 + , Pb 5 + ),	( K 3 + , I 3 + ),	( Ge 2 + , Sc 1 + ),
( Ti 3 + , Ar 1 + ),	( K 3 + , Hf 4 + ),	( Zr 3 + , Sc 1 + ),
( Se 3 + , Ar 1 + ),	( Bi 5 + , K 4 + ),	( Ag 2 + , Sc 1 + ),
( Sr 2 + , Ar 1 + ),	( Sc 5 + , K 5 + ),	( Hg 2 + , Sc 1 + ),
( Sb 3 + , Ar 1 + ),	( K 4 + , Fe 4 + ),	( Rb 2 + , Sc 2 + ),
( Gd 3 + , Ar 1 + ),	( K 4 + , Ni 4 + ),	( Sn 3 + , Sc 2 + ),
( Yb 3 + , Ar 1 + ),	( K 4 + , Cu 4 + ),	( Nd 3 + , Sc 2 + ),
( Fe 3 + , Ar 2 + ),	( Kr 6 + , K 5 + ),	( Pm 3 + , Sc 2 + ),
( Ni 3 + , Ar 2 + ),	( Ca 6 + , K 6 + ),	( Kr 3 + , Sc 3 + ),
( Cu 3 + , Ar 2 + ),	( V 5 + , K 6 + ),	( Rb 3 + , Sc 3 + ),
( Sb 4 + , Ar 2 + ),	( K 5 + , Mn 5 + ),	( Sc 3 + , Ge 4 + ),
( Bi 4 + , Ar 2 + ),	( As 5 + , K 6 + ),	( Sc 3 + , Mo 4 + ),
( Ar 2 + , Sc 2 + ),	( K 5 + , Sr 5 + ),	( Sc 3 + , Lu 4 + ),
( Ar 2 + , Ti 2 + ),	( K 5 + , Sn 5 + ),	( Sc 3 + , Bi 4 + ),
( Se 4 + , Ar 3 + ),	( K 7 + , Ca 7 + ),	( Ti 5 + , Sc 5 + ),
( Ar 2 + , Zr 2 + ),	( K 7 + , As 6 + ),	( Mn 6 + , Sc 5 + ),
( Mo 5 + , Ar 3 + ),	( K 7 + , Mo 7 + ),	( Sc 4 + , Ga 4 + ),
( Pb 4 + , Ar 3 + ),	( Mn 2 + , Ca 1 + ),	( Sc 4 + , As 5 + ),

( Ar 3 + , K 2 + ),	( Co 2 , Ca 1 + ),	( Cu 6 + , Sc 6 + ),
( Ar 3 + , Xe 3 + ),	( Ge 2 + , Ca 1 + ),	( Cu 7 + , Sc 7 + ),
( Ar 3 + , Pb 3 + ),	( Zr 3 + , Ca 1 + ),	( Ni 2 + , Ti 1 + ),
( Bi 5 + , Ar 4 + ),	( Hf 3 + , Ca 1 + ),	( Ge 2 + , Ti 1 + ),
( Ar 4 + , V 4 + ),	( Hg 2 + , Ca 1 + ),	( Zr 3 + , Ti 1 + ),
( Cu 5 + , Ar 5 + ),	( Zn 2 + , Ca 2 + ),	( Ag 2 + , Ti 1 + ),
( Ar 4 + , Br 4 + ),	( Rb 2 + , Ca 2 + ),	( Hg 2 + , Ti 1 + ),
( Sn 3 + , Ti 2 + ),	( Se 6 + , V 6 + ),	( Mn 2 + , Dy 1 + ),
( Pm 3 + , Ti 2 + ),	( V 6 + , Sr 8 + ),	( Mn 2 + , Ho 1 + ),
( Sm 3 + , Ti 2 + ),	( Ni 2 + , Cr 1 + ),	( Mn 2 + , Er 1 + ),
( Dy 3 + , Ti 2 + ),	( Ge 2 + , Cr 1 + ),	( Mn 2 + , Tm 1 + ),
( Fe 3 + , Ti 3 + ),	( Zr 3 + , Cr 1 + ),	( Mn 2 + , Yb 1 + ),
( Ni 3 + , Ti 3 + ),	( Ag 2 + , Cr 1 + ),	( Mn 2 + , Lu 1 + ),
( Cu 3 + , Ti 3 + ),	( Hg 2 + , Cr 1 + ),	( Mn 2 + , Hf 1 + ),
( Ti 3 + , Mn 2 + ),	( Sr 2 + , Cr 2 + ),	( Mn 2 + , Tl 1 + ),
( Ti 3 + , Fe 2 + ),	( Sb 3 + , Cr 2 + ),	( Mn 2 + , Ra 1 + ),
( Ti 3 + , Ge 2 + ),	( Gd 3 + , Cr 2 + ),	( Mn 2 + , Ac 1 + ),
( Rb 4 + , Ti 4 + ),	( Yb 3 + , Cr 2 + ),	( Mn 2 + , Th 1 + ),
( Sr 4 + , Ti 4 + ),	( Zn 3 + , Cr 3 + ),	( Mn 2 + , Pa 1 + ),
( Ti 3 + , Mo 2 + ),	( Te 4 + , Cr 3 + ),	( Mn 2 + , U 1 + ),
( Ti 3 + , Tc 2 + ),	( Cr 2 + , Cs 1 + ),	( Mn 2 + , Np 1 + ),
( Te 5 + , Ti 4 + ),	( Cr 3 + , Se 2 + ),	( Mn 2 + , Pu 1 + ),
( Ti 3 + , Hf 2 + ),	( Cr 3 + , Br 2 + ),	( Mn 2 + , Am 1 + ),
( Ti 3 + , Pb 2 + ),	( Y 4 + , Cr 4 + ),	( Mn 2 + , Cm 1 + ),
( As 5 + , Ti 5 + ),	( Cr 3 + , Ag 2 + ),	( Mn 2 + , Bk 1 + ),
( Ti 4 + , Rb 5 + ),	( Cr 3 + , Xe 2 + ),	( Mn 2 + , Cf 1 + ),
( Ti 4 + , Sr 5 + ),	( Cr 3 + , Pr 3 + ),	( Mn 2 + , Es 1 + ),
( Mo 6 + , Ti 5 + ),	( Cr 3 + , Gd 3 + ),	( Co 4 + , Mn 4 + ),
( Ti 7 + , Ti 7 + ),	( Cr 3 + , Tb 3 + ),	( Kr 5 + , Mn 4 + ),
( Ti 7 + , Ti 7 + ),	( Cr 3 + , Lu 3 + ),	( Mn 3 + , Zr 3 + ),
( Mn 7 + , Ti 8 + ),	( Cr 4 + , Pm 4 + ),	( Mn 3 + , Sm 3 + ),
( Ni 2 + , V 1 + ),	( Cr 4 + , Sm 4 + ),	( Mn 3 + , Dy 3 + ),
( Ge 2 + , V 1 + ),	( Cr 4 + , Dy 4 + ),	( Mn 3 + , Ho 3 + ),
( Zr 3 + , V 1 + ),	( Cr 6 + , Ni 7 + ),	( Mn 3 + , Er 3 + ),
( Ag 2 + , V 1 + ),	( Cr 6 + , Zn 7 + ),	( Mn 3 + , Tm 3 + ),
( Hg 2 + , V 1 + ),	( Cr 7 + , Co 8 + ),	( Mn 3 + , Hf 3 + ),
( Se 3 + , V 2 + ),	( Ni 2 + , Mn 1 + ),	( Mn 4 + , Sb 4 + ),
( Eu 3 + , V 2 + ),	( Ag 2 + , Mn 1 + ),	( Mn 4 + , Gd 4 + ),
( Ho 3 + , V 2 + ),	( Se 3 + , Mn 2 + ),	( Mn 4 + , Lu 4 + ),
( Er 3 + , V 2 + ),	( Sr 2 + , Mn 2 + ),	( Mn 4 + , Bi 4 + ),
( Tm 3 + , V 2 + ),	( Gd 3 + , Mn 2 + ),	( Sr 7 + , Mn 6 + ),
( Pb 3 + , V 2 + ),	( Tm 3 + , Mn 2 + ),	( Mn 6 + , Sr 6 + ),
( Sr 3 + , V 3 + ),	( Yb 3 + , Mn 2 + ),	( Ni 2 + , Fe 1 + ),
( Fe 4 + , V 4 + ),	( Mn 2 + , Ga 1 + ),	( Br 2 + , Fe 1 + ),
( V 3 + , As 2 + ),	( Mn 2 + , Sr 1 + ),	( Sr 2 + , Fe 2 + ),
( V 3 + , Pd 2 + ),	( Mn 2 + , Y 1 + ),	( Sb 3 + , Fe 2 + ),
( V 3 + , In 2 + ),	( Y 3 + , Mn 3 + ),	( Gd 3 + , Fe 2 + ),
( V 3 + , Te 2 + ),	( Mo 4 + , Mn 3 + ),	( Yb 3 + , Fe 2 + ),
( V 3 + , I 2 + ),	( Mn 2 + , In 1 + ),	( Te 4 + , Fe 3 + ),

( V 3 + , La 3 + ),	( Mn 2 + , Ba 1 + ),	( Zn 4 + , Fe 4 + ),
( V 3 + , Pt 2 + ),	( Mn 2 + , La 1 + ),	( Fe 3 + , Rb 2 + ),
( V 3 + , Hg 2 + ),	( Mn 2 + , Ce 1 + ),	( Fe 3 + , Mo 3 + ),
( V 4 + , Cu 3 + ),	( Mn 2 + , Pr 1 + ),	( Cu 5 + , Fe 5 + ),
( Ge 4 + , V 5 + ),	( Mn 2 + , Nd 1 + ),	( Fe 4 + , Br 4 + ),
( V 4 + , Kr 3 + ),	( Mn 2 + , Pm 1 + ),	( Br 6 + , Fe 5 + ),
( Y 5 + , V 5 + ),	( Mn 2 + , Sm 1 + ),	( Nb 5 + , Fe 5 + ),
( V 4 + , Cd 3 + ),	( Mn 2 + , Eu 1 + ),	( Fe 5 + , Rb 5 + ),
( V 4 + , Te 4 + ),	( Mn 2 + , Gd 1 + ),	( Fe 5 + , Sr 5 + ),
( V 4 + , Ce 4 + ),	( Mn 2 + , Tb 1 + ),	( Mo 6 + , Fe 6 + ),
( Fe 5 + , Te 6 + ),	( Co 7 + , Y 8 + ),	( Zn 4 + , Cu 4 + ),
( Mo 7 + , Fe 7 + ),	( Ni 2 + , Ni 1 + ),	( Cu 3 + , Rb 2 + ),
( Ni 2 + , Co 1 + ),	( Br 2 + , Ni 1 + ),	( Cu 3 + , Mo 3 + ),
( Br 2 + , Co 1 + ),	( Ag 2 + , Ni 1 + ),	( Cu 3 + , In 3 + ),
( Sb 3 + , Co 2 + ),	( Ge 3 + , Ni 2 + ),	( Cu 3 + , Te 3 + ),
( Lu 3 + , Co 2 + ),	( Mo 3 + , Ni 2 + ),	( Zn 5 + , Cu 5 + ),
( Bi 3 + , Co 2 + ),	( Lu 3 + , Ni 2 + ),	( Cu 4 + , Kr 4 + ),
( Co 2 + , Ga 1 + ),	( Bi 3 + , Ni 2 + ),	( Cu 4 + , Rb 4 + ),
( Co 2 + , Sr 1 + ),	( Ni 2 + , Ni 1 + ),	( Sb 5 + , Cu 5 + ),
( Co 2 + , Y 1 + ),	( Ni 2 + , Cu 1 + ),	( Cu 6 + , Kr 7 + ),
( Y 3 + , Co 3 + ),	( Ni 2 + , Ge 1 + ),	( Kr 2 + , Zn 1 + ),
( Mo 4 + , Co 3 + ),	( As 4 + , Ni 3 + ),	( Cd 2 + , Zn 1 + ),
( Co 2 + , In 1 + ),	( Ni 2 + , Zr 1 + ),	( Te 3 + , Zn 1 + ),
( Co 2 + , Ba 1 + ),	( Ni 2 + , Nb 1 + ),	( Ce 3 + , Zn 1 + ),
( Co 2 + , La 1 + ),	( Ni 2 + , Mo 1 + ),	( Ge 3 + , Zn 2 + ),
( Co 2 + , Ce 1 + ),	( Ni 2 + , Tc 1 + ),	( Mo 3 + , Zn 2 + ),
( Co 2 + , Pr 1 + ),	( Ni 2 + , Ru 1 + ),	( Lu 3 + , Zn 2 + ),
( Co 2 + , Nd 1 + ),	( Ni 2 + , Rh 1 + ),	( Bi 3 + , Zn 2 + ),
( Co 2 + , Pm 1 + ),	( Ni 2 + , Ag 1 + ),	( Zn 2 + , Br 1 + ),
( Co 2 + , Sm 1 + ),	( Ni 2 + , Sn 1 + ),	( Zn 2 + , Y 2 + ),
( Co 2 + , Eu 1 + ),	( Ni 2 + , Ta 1 + ),	( Mo 5 + , Zn 3 + ),
( Co 2 + , Gd 1 + ),	( Ni 2 + , W 1 + ),	( Zn 2 + , Xe 1 + ),
( Co 2 + , Tb 1 + ),	( Ni 2 + , Re 1 + ),	( Zn 2 + , Eu 2 + ),
( Co 2 + , Dy 1 + ),	( Ni 2 + , Pb 1 + ),	( Zn 2 + , Gd 2 + ),
( Co 2 + , Ho 1 + ),	( Ni 2 + , Bi 1 + ),	( Zn 2 + , Tb 2 + ),
( Co 2 + , Er 1 + ),	( Zn 4 + , Ni 4 + ),	( Zn 2 + , Dy 2 + ),
( Co 2 + , Tm 1 + ),	( Ni 3 + , Rb 2 + ),	( Zn 2 + , Ho 2 + ),
( Co 2 + , Yb 1 + ),	( Ni 3 , Mo 3 + ),	( Zn 2 + , Er 2 + ),
( Co 2 + , Lu 1 + ),	( Cu 5 + , Ni 5 + ),	( Zn 2 + , Tm 2 + ),
( Co 2 + , Tl 1 + ),	( Ni 4 + , Br 4 + ),	( Zn 2 + , Yb 2 + ),
( Co 2 + , Ra 1 + ),	( Br 6 + , Ni 5 + ),	( Zn 3 + , Rh 3 + ),
( Co 2 + , Ac 1 + ),	( Nb 5 + , Ni 5 + ),	( Zn 3 + , Xe 3 + ),
( Co 2 + , Th 1 + ),	( Ni 5 + , Cu 5 + ),	( Zn 3 + , Pb 3 + ),
( Co 2 + , Pa 1 + ),	( Rb 7 + , Ni 6 + ),	( Kr 6 + , Zn 5 + ),
( Co 2 + , U 1 + ),	( Ni 7 + , Zn 7 + ),	( Rb 7 + , Zn 6 + ),
( Co 2 + , Np 1 + ),	( Br 2 + , Cu 1 + ),	( Zn 6 + , Sr 7 + ),
( Co 2 + , Pu 1 + ),	( Ag 2 + , Cu 1 + ),	( Ge 2 + , Ga 1 + ),
( Co 2 + , Am 1 + ),	( Br 3 + , Cu 2 + ),	( Zr 3 + , Ga 1 + ),
( Co 2 + , Cm 1 + ),	( Cu 2 + , Zn 1 + ),	( I 2 + , Ga 1 + ),

( Co 2 + , Bk 1 + ),	( Ga 3 + , Cu 3 + ),	( Hf 3 + , Ga 1 + ),
( Co 2 + , Cf 1 + ),	( Cu 2 + , As 1 + ),	( Hg 2 + , Ga 1 + ),
( Co 2 + , Es 1 + ),	( Cu 2 + , Se 1 + ),	( Te 4 + , Ga 3 + ),
( Co 4 + , Co 4 + ),	( Kr 4 + , Cu 3 + ),	( Ga 3 + , Br 3 + ),
( Kr 5 + , Co 4 + ),	( Cu 2 + , Pd 1 + ),	( Ga 3 + , Kr 3 + ),
( Co 3 + , Zr 3 + ),	( Cu 2 + , Cd 1 + ),	( Ga 3 + , Ce 4 + ),
( Co 3 + , Sm 3 + ),	( Cu 2 + , Sb 1 + ),	( Br 2 + , Ge 1 + ),
( Co 3 + , Ho 3 + ),	( Cu 2 + , Te 1 + ),	( Se 3 + , Ge 2 + ),
( Co 3 + , Tm 3 + ),	( Cu 2 + , Os 1 + ),	( Sr 2 + , Ge 2 + ),
( Co 3 + , Hf 3 + ),	( Cu 2 + , Ir 1 + ),	( Sb 3 + , Ge 2 + ),
( Co 4 + , Co 4 + ),	( Cu 2 + , Pt 1 + ),	( Gd 3 + , Ge 2 + ),
( Co 7 + , Co 7 + ),	( Cu 2 + , Au 1 + ),	( Yb 3 + , Ge 2 + ),
( Co 7 + , Co 7 + ),	( Cu 2 + , Po 1 + ),	( Ge 2 + , Y 1 + ),
( Y 3 + , Ge 3 + ),	( Te 4 + , Se 3 + ),	( Kr 3 + , Eu 3 + ),
( Ge 2 + , Zr 1 + ),	( Rb 4 + , Se 4 + ),	( Kr 3 + , Yb 3 + ),
( Ge 2 + , Nb 1 + ),	( Se 3 + , Tc 2 + ),	( Kr 4 + , Kr 3 + ),
( Ge 2 + , Mo 1 + ),	( Se 3 + , Sn 2 + ),	( Y 5 + , Kr 5 + ),
( Ge 2 + , In 1 + ),	( Te 5 + , Se 4 + ),	( Kr 4 + , Cd 3 + ),
( Ge 2 + , Gd 1 + ),	( Se 3 + , Hf 2 + ),	( Kr 4 + , Te 4 + ),
( Ge 2 + , Tb 1 + ),	( Se 3 + , Pb 2 + ),	( Kr 4 + , Ce 4 + ),
( Ge 2 + , Dy 1 + ),	( Se 4 + , Rb 3 + ),	( Sr 6 + , Kr 6 + ),
( Ge 2 + , Ho 1 + ),	( Se 4 + , Sn 4 + ),	( Kr 5 + , Nb 5 + ),
( Ge 2 + , Er 1 + ),	( Se 4 + , Nd 4 + ),	( Xe 2 + , Rb 1 + ),
( Ge 2 + , Tm 1 + ),	( Se 4 + , Pm 4 + ),	( Pb 2 + , Rb 1 + ),
( Ge 2 + , Yb 1 + ),	( Se 5 + , In 4 + ),	( Rb 2 + , Y 2 + ),
( Ge 2 + , Hf 1 + ),	( Rb 2 + , Br 1 + ),	( Mo 5 + , Rb 3 + ),
( Ge 2 + , Tl 1 + ),	( Pr 3 + , Br 1 + ),	( Rb 2 + , Xe 1 + ),
( Ge 2 + , Th 1 + ),	( Tb 3 + , Br 1 + ),	( Rb 2 + , Gd 2 + ),
( Ge 2 + , Pa 1 + ),	( La 3 + , Br 2 + ),	( Rb 2 + , Tb 2 + ),
( Ge 2 + , U 1 + ),	( Br 2 + , Pd 1 + ),	( Rb 2 + , Dy 2 + ),
( Ge 2 + , Np 1 + ),	( Br 2 + , Ag 1 + ),	( Rb 2 + , Ho 2 + ),
( Ge 2 + , Pu 1 + ),	( Br 2 + , Cd 1 + ),	( Rb 2 + , Er 2 + ),
( Ge 2 + , Am 1 + ),	( Br 2 + , Sb 1 + ),	( Rb 2 + , Tm 2 + ),
( Ge 2 + , Cm 1 + ),	( Br 2 + , Ta 1 + ),	( Rb 2 + , Yb 2 + ),
( Ge 2 + , Bk 1 + ),	( Br 2 + , W 1 + ),	( Rb 3 + , Nb 3 + ),
( Ge 2 + , Cf 1 + ),	( Br 2 + , Re 1 + ),	( Rb 3 + , Sb 3 + ),
( Ge 2 + , Es 1 + ),	( Br 2 + , Os 1 + ),	( Rb 3 + , Cs 2 + ),
( Ge 3 + , As 2 + ),	( Br 2 + , Po 1 + ),	( Rb 3 + , Eu 3 + ),
( Ge 3 + , Rh 2 + ),	( Br 3 + , Pd 2 + ),	( Rb 3 + , Yb 3 + ),
( Ge 3 + , Te 2 + ),	( Br 3 + , In 2 + ),	( Rb 3 + , Bi 3 + ),
( Ge 3 + , Pt 2 + ),	( Br 3 + , I 2 + ),	( Rb 6 + , Rb 5 + ),
( Kr 2 + , As 1 + ),	( Br 3 + , La 3 + ),	( Rb 4 + , Sr 3 + ),
( Nb 3 + , As 1 + ),	( Br 3 + , Ce 3 + ),	( Rb 4 + , Eu 4 + ),
( Cd 2 + , As 1 + ),	( Br 4 + , Xe 3 + ),	( Rb 4 + , Er 4 + ),
( Te 3 + , As 1 + ),	( Br 4 + , Pb 3 + ),	( Rb 4 + , Tm 4 + ),
( Mo 3 + , As 2 + ),	( Y 6 + , Br 6 + ),	( Rb 4 + , Yb 4 + ),
( Sb 4 + , As 3 + ),	( Br 5 + , Mo 5 + ),	( Rb 5 + , Sr 4 + ),
( Bi 4 + , As 3 + ),	( Pm 3 + , Kr 1 + ),	( Rb 5 + , Sb 5 + ),
( As 3 + , Br 2 + ),	( Sm 3 + , Kr 1 + ),	( Rb 5 + , Bi 5 + ),



( Kr 5 + , As 4 + ),	( Dy 3 + , Kr 1 + ),	( Rb 6 + , Rb 5 + ),
( As 3 + , Zr 3 + ),	( Pb 3 + , Kr 1 + ),	( Rb 6 + , Sr 5 + ),
( As 3 + , Nd 3 + ),	( Kr 3 + , Kr 2 + ),	( Mo 6 + , Rb 7 + ),
( As 3 + , Pm 3 + ),	( Rb 3 + , Kr 2 + ),	( Rb 7 + , Sb 6 + ),
( As 3 + , Tb 3 + ),	( Kr 4 + , Kr 3 + ),	( Pd 2 + , Sr 1 + ),
( As 3 + , Dy 3 + ),	( Kr 2 + , Cd 1 + ),	( I 2 + , Sr 1 + ),
( As 3 + , Ho 3 + ),	( Kr 2 + , Sb 1 + ),	( Hf 3 + , Sr 1 + ),
( As 3 + , Er 3 + ),	( Kr 2 + , Te 1 + ),	( Nb 3 + , Sr 2 + ),
( As 4 + , Br 3 + ),	( Kr 2 + , Os 1 + ),	( Pr 3 + , Sr 2 + ),
( Sr 5 + , As 5 + ),	( Kr 2 + , Ir 1 + ),	( Sr 4 + , Sr 3 + ),
( Se 6 + , As 6 + ),	( Kr 2 + , Pt 1 + ),	( Sr 2 + , Mo 2 + ),
( As 5 + , Rb 7 + ),	( Kr 2 + , Au 1 + ),	( Sr 2 + , Tc 2 + ),
( Kr 2 + , Se 1 + ),	( Kr 3 + , Kr 2 + ),	( Sr 2 + , Sb 2 + ),
( Cd 2 + , Se 1 + ),	( Kr 3 + , Nb 3 + ),	( Te 5 + , Sr 3 + ),
( Te 3 + , Se 1 + ),	( Kr 3 + , Sb 3 + ),	( Sr 3 + , Tc 3 + ),
( Ce 3 + , Se 1 + ),	( Kr 3 + , Cs 2 + ),	( Sr 3 + , Tl 3 + ),
( Sr 4 + , Sr 3 + ),	( Eu 3 + , Nb 2 + ),	( Ag 2 + , Ru 1 + ),
( Sr 4 + , Sb 4 + ),	( Dy 3 + , Nb 2 + ),	( Sb 3 + , Ru 2 + ),
( Sr 4 + , Gd 4 + ),	( Ho 3 + , Nb 2 + ),	( Gd 3 + , Ru 2 + ),
( Sr 4 + , Yb 4 + ),	( Er 3 + , Nb 2 + ),	( Lu 3 + , Ru 2 + ),
( Zr 3 + , Y 1 + ),	( Tm 3 + , Nb 2 + ),	( Sb 4 + , Ru 3 + ),
( Ag 2 + , Y 1 + ),	( Pb 3 + , Nb 2 + ),	( Bi 4 + , Ru 3 + ),
( Hg 2 + , Y 1 + ),	( Nb 3 + , I 1 + ),	( Ag 2 + , Rh 1 + ),
( Sn 3 + , Y 2 + ),	( Nb 3 + , Ba 2 + ),	( Lu 3 + , Rh 2 + ),
( Nd 3 + , Y 2 + ),	( Nb 3 + , La 2 + ),	( Bi 3 + , Rh 2 + ),
( Tb 3 + , Y 2 + ),	( Nb 3 + , Ce 2 + ),	( Te 4 + , Rh 3 + ),
( Y 3 + , Zr 4 + ),	( Nb 3 + , Pr 2 + ),	( Rh 2 + , Cs 1 + ),
( Y 3 + , Hf 4 + ),	( Nb 3 + , Nd 2 + ),	( Ce 3 + , Pd 1 + ),
( Y 3 + , Hg 3 + ),	( Nb 3 + , Pm 2 + ),	( Pd 2 + , In 1 + ),
( Y 4 + , La 4 + ),	( Nb 3 + , Sm 2 + ),	( Pd 2 + , Ba 1 + ),
( Y 6 + , Bi 6 + ),	( Nb 3 + , Eu 2 + ),	( Pd 2 + , La 1 + ),
( Zr 3 + , Zr 1 + ),	( Nb 3 + , Hg 1 + ),	( Pd 2 + , Ce 1 + ),
( Ag 2 + , Zr 1 + ),	( Nb 3 + , Rn 1 + ),	( Pd 2 + , Pr 1 + ),
( Hg 2 + , Zr 1 + ),	( Nb 3 + , Ra 2 + ),	( Pd 2 + , Nd 1 + ),
( Sn 3 + , Zr 2 + ),	( Nb 4 + , Nd 3 + ),	( Pd 2 + , Pm 1 + ),
( Nd 3 + , Zr 2 + ),	( Nb 4 + , Pm 3 + ),	( Pd 2 + , Sm 1 + ),
( Pm 3 + , Zr 2 + ),	( Nb 4 + , Sm 3 + ),	( Pd 2 + , Eu 1 + ),
( Sm 3 + , Zr 2 + ),	( Nb 4 + , Dy 3 + ),	( Pd 2 + , Tb 1 + ),
( Dy 3 + , Zr 2 + ),	( Nb 4 + , Ho 3 + ),	( Pd 2 + , Dy 1 + ),
( Nb 4 + , Zr 3 + ),	( Nb 4 + , Er 3 + ),	( Pd 2 + , Lu 1 + ),
( Zr 3 + , Zr 1 + ),	( Nb 4 + , Hf 3 + ),	( Pd 2 + , Ra 1 + ),
( Zr 3 + , Nb 1 + ),	( Mo 7 + , Nb 7 + ),	( Pd 2 + , Ac 1 + ),
( Zr 3 + , Mo 1 + ),	( Ag 2 + , Mo 1 + ),	( Pd 2 + , Pa 1 + ),
( Zr 3 + , Tc 1 + ),	( Hg 2 + , Mo 1 + ),	( Ag 2 + , Ag 1 + ),
( Zr 3 + , Gd 1 + ),	( Sb 3 + , Mo 2 + ),	( La 3 + , Ag 2 + ),
( Zr 3 + , Tb 1 + ),	( Gd 3 + , Mo 2 + ),	( Ag 2 + , Ag 1 + ),
( Zr 3 + , Dy 1 + ),	( Yb 3 + , Mo 2 + ),	( Ag 2 + , Sn 1 + ),
( Zr 3 + , Ho 1 + ),	( Mo 3 + , Rh 2 + ),	( Ag 2 + , Hf 1 + ),
( Zr 3 + , Er 1 + ),	( Mo 3 + , In 2 + ),	( Ag 2 + , Pb 1 + ),

( Zr 3 + , Tm 1 + ),	( Mo 3 + , Te 2 + ),	( Ag 2 + , Bi 1 + ),
( Zr 3 + , Yb 1 + ),	( Mo 3 + , I 2 + ),	( Ag 2 + , Es 1 + ),
( Zr 3 + , Hf 1 + ),	( Mo 3 + , La 3 + ),	( Cd 2 + , Cd 1 + ),
( Zr 3 + , Tl 1 + ),	( Mo 3 + , Pt 2 + ),	( Te 3 + , Cd 1 + ),
( Zr 3 + , Bi 1 + ),	( Mo 3 + , Hg 2 + ),	( Ce 3 + , Cd 1 + ),
( Zr 3 + , Th 1 + ),	( Mo 4 + , Pd 3 + ),	( Sb 3 + , Cd 2 + ),
( Zr 3 + , Pa 1 + ),	( Mo 4 + , I 3 + ),	( Gd 3 + , Cd 2 + ),
( Zr 3 + , U 1 + ),	( Mo 4 + , Hf 4 + ),	( Lu 3 + , Cd 2 + ),
( Zr 3 + , Np 1 + ),	( Bi 5 + , Mo 5 + ),	( Bi 3 + , Cd 2 + ),
( Zr 3 + , Pu 1 + ),	( Mo 5 + , Sn 4 + ),	( Cd 2 + , Cd 1 + ),
( Zr 3 + , Am 1 + ),	( Mo 5 + , Nd 4 + ),	( Cd 2 + , Te 1 + ),
( Zr 3 + , Cm 1 + ),	( Mo 5 + , Tb 4 + ),	( Cd 2 + , I 1 + ),
( Zr 3 + , Bk 1 + ),	( Ag 2 + , Tc 1 + ),	( Cd 2 + , Ba 2 + ),
( Zr 3 + , Cf 1 + ),	( Eu 3 + , Tc 2 + ),	( Cd 2 + , Ir 1 + ),
( Zr 3 + , Es 1 + ),	( Ho 3 + , Tc 2 + ),	( Cd 2 + , Pt 1 + ),
( Zr 4 + , In 4 + ),	( Er 3 + , Tc 2 + ),	( Cd 2 + , Au 1 + ),
( Ag 2 + , Nb 1 + ),	( Tm 3 + , Tc 2 + ),	( Cd 2 + , Hg 1 + ),
( Hg 2 + , Nb 1 + ),	( Yb 3 + , Tc 2 + ),	( Cd 2 + , Ra 2 + ),
( Sm 3 + , Nb 2 + ),	( Pb 3 + , Tc 2 + ),	( I 2 + , In 1 + ),
( Hf 3 + , In 1 + ),	( Tb 3 + , Xe 1 + ),	( Hg 2 + , Tb 1 + ),
( Hg 2 + , In 1 + ),	( Xe 2 + , Cs 1 + ),	( Tb 3 + , Tb 2 + ),
( Sb 4 + , In 3 + ),	( Pb 2 + , Cs 1 + ),	( Tb 3 + , Tb 2 + ),
( Bi 4 + , In 3 + ),	( Hf 3 + , Ba 1 + ),	( Tb 3 + , Dy 2 + ),
( In 3 + , Bi 3 + ),	( Hf 3 + , La 1 + ),	( Tb 3 + , Ho 2 + ),
( Eu 3 + , Sn 2 + ),	( Pr 3 + , La 2 + ),	( Tb 3 + , Er 2 + ),
( Ho 3 + , Sn 2 + ),	( La 3 + , Pr 3 + ),	( Tb 3 + , Tm 2 + ),
( Er 3 + , Sn 2 + ),	( La 3 + , Nd 3 + ),	( Tb 3 + , Yb 2 + ),
( Tm 3 + , Sn 2 + ),	( La 3 + , Pm 3 + ),	( Hf 3 + , Dy 1 + ),
( Pb 3 + , Sn 2 + ),	( La 3 + , Tb 3 + ),	( Hg 2 + , Dy 1 + ),
( Te 4 + , Sn 3 + ),	( La 3 + , Dy 3 + ),	( Dy 3 + , Lu 2 + ),
( Pb 4 + , Sn 4 + ),	( La 3 + , Ho 3 + ),	( Pb 4 + , Dy 4 + ),
( Sn 4 + , Sb 4 + ),	( La 3 + , Er 3 + ),	( Hf 3 + , Ho 1 + ),
( Sn 4 + , Gd 4 + ),	( Hf 3 + , Ce 1 + ),	( Hg 2 + , Ho 1 + ),
( Sn 4 + , Lu 4 + ),	( Pr 3 + , Ce 2 + ),	( Ho 3 + , Hf 2 + ),
( Ce 3 + , Sb 1 + ),	( Ce 3 + , Os 1 + ),	( Ho 3 + , Pb 2 + ),
( Sb 3 + , Sb 2 + ),	( Ce 3 + , Ir 1 + ),	( Hf 3 + , Er 1 + ),
( Gd 3 + , Sb 2 + ),	( Ce 3 + , Pt 1 + ),	( Hg 2 + , Er 1 + ),
( Yb 3 + , Sb 2 + ),	( Ce 3 + , Au 1 + ),	( Er 3 + , Hf 2 + ),
( Sb 3 + , Sb 2 + ),	( Ce 3 + , Po 1 + ),	( Er 3 + , Pb 2 + ),
( Sb 3 + , Bi 2 + ),	( Hf 3 + , Pr 1 + ),	( Hf 3 + , Tm 1 + ),
( Sb 4 + , Te 3 + ),	( Pr 3 + , Pr 2 + ),	( Hg 2 + , Tm 1 + ),
( Te 3 + , Te 1 + ),	( Pr 3 + , Pr 2 + ),	( Tm 3 + , Hf 2 + ),
( Ce 3 + , Te 1 + ),	( Pr 3 + , Nd 2 + ),	( Tm 3 + , Pb 2 + ),
( Bi 4 + , Te 3 + ),	( Pr 3 + , Pm 2 + ),	( Hf 3 + , Yb 1 + ),
( Te 3 + , Te 1 + ),	( Pr 3 + , Sm 2 + ),	( Hg 2 + , Yb 1 + ),
( Te 3 + , Ba 2 + ),	( Pr 3 + , Eu 2 + ),	( Yb 3 + , Bi 2 + ),
( Te 3 + , Ir 1 + ),	( Pr 3 + , Tb 2 + ),	( Hf 3 + , Lu 1 + ),
( Te 3 + , Pt 1 + ),	( Pr 3 + , Dy 2 + ),	( Pb 3 + , Lu 2 + ),
( Te 3 + , Au 1 + ),	( Pr 3 + , Ho 2 + ),	( Lu 3 + , Bi 2 + ),

( Te 3 + , Ra 2 + ),	( Pr 3 + , Er 2 + ),	( Hg 2 + , Hf 1 + ),
( Te 5 + , Eu 4 + ),	( Pr 3 + , Rn 1 + ),	( Pb 3 + , Hf 2 + ),
( Te 5 + , Ho 4 + ),	( Hf 3 + , Nd 1 + ),	( Hf 3 + , Tl 1 + ),
( Te 5 + , Er 4 + ),	( Nd 3 + , Gd 2 + ),	( Hf 3 + , Ra 1 + ),
( Te 5 + , Tm 4 + ),	( Nd 3 + , Er 2 + ),	( Hf 3 + , Ac 1 + ),
( Te 5 + , Pb 4 + ),	( Nd 3 + , Tm 2 + ),	( Hf 3 + , Th 1 + ),
( I 2 + , Ba 1 + ),	( Nd 3 + , Yb 2 + ),	( Hf 3 + , Pa 1 + ),
( I 2 + , La 1 + ),	( Pb 4 + , Nd 4 + ),	( Hf 3 + , U 1 + ),
( I 2 + , Ce 1 + ),	( Hf 3 + , Pm 1 + ),	( Hf 3 + , Np 1 + ),
( I 2 + , Pr 1 + ),	( Pm 3 + , Lu 2 + ),	( Hf 3 + , Pu 1 + ),
( I 2 + , Nd 1 + ),	( Pb 4 + , Pm 4 + ),	( Hf 3 + , Am 1 + ),
( I 2 + , Pm 1 + ),	( Hf 3 + , Sm 1 + ),	( Hf 3 + , Cm 1 + ),
( I 2 + , Sm 1 + ),	( Sm 3 + , Lu 2 + ),	( Hf 3 + , Bk 1 + ),
( I 2 + , Eu 1 + ),	( Pb 4 + , Sm 4 + ),	( Hf 3 + , Cf 1 + ),
( I 2 + , Tb 1 + ),	( Hf 3 + , Eu 1 + ),	( Hg 2 + , Tl 1 + ),
( I 2 + , Dy 1 + ),	( Eu 3 + , Hf 2 + ),	( Hg 2 + , Th 1 + ),
( I 2 + , Lu 1 + ),	( Eu 3 + , Pb 2 + ),	( Hg 2 + , Pa 1 + ),
( I 2 + , Ra 1 + ),	( Hf 3 + , Gd 1 + ),	( Hg 2 + , U 1 + ),
( I 2 + , Ac 1 + ),	( Hg 2 + , Gd 1 + ),	( Hg 2 + , Np 1 + ),
( I 2 + , Pa 1 + ),	( Tb 3 + , Gd 2 + ),	( Hg 2 + , Pu 1 + ),
( I 2 + , Am 1 + ),	( Gd 3 + , Bi 2 + ),	( Hg 2 + , Am 1 + ),
( Nd 3 + , Xe 1 + ),	( Hf 3 + , Tb 1 + ),	( Hg 2 + , Cm 1 + ),
( Hg 2 + , Bk 1 + ),	( Hg 2 + , Cf 1 + ),	( Hg 2 + , Es 1 + ),
( Pb 3 + , Pb 2 + ),	( Pb 3 + , Pb 2 + ),	( K 1 + , Cl ),
( As 2 + , H ),	( K 1 + , F ),	( Cr 2 + , Cl ),
( Ru 2 + , H ),	( Cr 2 + , F ),	( Fe 2 + , Cl ),
( In 2 + , H ),	( Fe 2 , F ),	( As 2 + , K ),
( Te 2 + , H ),	( As 2 + , Na ),	( Ru 2 + , K ),
( Al 2 + , H ),	( Ru 2 + , Na ),	( In 2 + , K ),
( Ar 1 + , H ),	( In 2 + , Na ),	( Te 2 + , K ),
( As 2 + , Li ),	( Te 2 + , Na ),	( Al 2 + , K ),
( Ru 2 + , Li ),	( Al 2 + , Na ),	( Ar 1 + , K ),
( In 2 + , Li ),	( Ar 1 + , Na ),	( As 2 + , Fe ),
( Te 2 + , Li ),	( Ti 2 + , Na ),	( Ru 2 + , Fe ),
( Al 2 + , Li ),	( As 2 + , Al ),	( In 2 + , Fe ),
( Ar 1 + , Li ),	( Ru 2 + , Al ),	( Te 2 + , Fe ),
( Ti 2 + , Li ),	( In 2 + , Al ),	( Al 2 + , Fe ),
( As 2 + , B ),	( Te 2 + , Al ),	( Ar 1 + , Fe ),
( Rb 1 + , B ),	( Al 2 + , Al ),	( Ti 2 + , Fe ),
( Mo 2 + , B ),	( Ar 1 + , Al ),	( As 2 + , Co ),
( Ru 2 + , B ),	( Ti 2 + , Al ),	( Ru 2 + , Co ),
( In 2 + , B ),	( As 2 + , Si ),	( In 2 + , Co ),
( Te 2 + , B ),	( Tc 2 + , Si ),	( Te 2 + , Co ),
( Al 2 + , B ),	( Ru 2 + , Si ),	( Al 2 + , Co ),
( Ar 1 + , B ),	( Ti 2 + , Si ),	( V 2 + , Co ),
( Ti 2 + , B ),	( N 1 + , Si ),	( Tc 2 + , Cu ),
( As 2 + , C ),	( Al 2 + , Si ),	( Ti 2 + , Cu ),
( Tc 2 + , C ),	( V 2 + , Si ),	( N 1 + , Cu ),
( Ru 2 + , C ),	( As 2 + , P ),	( P 2 + , Cu ),

( In 2 + , C ),	( Ru 2 + , P ),	( V 2 + , Cu ),
( Te 2 + , C ),	( In 2 + , P ),	( Ga 2 + , Br ),
( N 1 + , C ),	( Te 2 + , P ),	( Se 2 + , Br ),
( Al 2 + , C ),	( Al 2 + , P ),	( Rh 2 + , Br ),
( V 2 + , C ),	( Ar 1 + , P ),	( Sn 2 + , Br ),
( As 2 + , O ),	( Tc 2 + , S ),	( P 2 + , Br ),
( Tc 2 + , O ),	( Sn 2 + , S ),	( K 1 + , Br ),
( Ru 2 + , O ),	( Tl 2 + , S ),	( Cr 2 + , Br ),
( Tl 2 + , O ),	( N 1 + , S ),	( Fe 2 + , Br ),
( N 1 + , O ),	( P 2 + , S ),	( As 2 + , Rb ),
( Al 2 + , O ),	( V 2 + , S ),	( Rb 1 + , Rb ),
( V 2 + , O ),	( Ga 2 + , Cl ),	( Mo 2 + , Rb ),
( Ga 2 + , F ),	( Se 2 + , Cl ),	( Ru 2 + , Rb ),
( Se 2 + , F ),	( Rh 2 + , Cl ),	( In 2 + , Rb ),
( Rh 2 + , F ),	( Sn 2 + , Cl ),	( Te 2 + , Rb ),
( Sn 2 + , F ),	( Xe 2 + , Cl ),	( Al 2 + , Rb ),
( Pb 2 + , F ),	( Pb 2 + , Cl ),	( Ru 2 + , Pb ),
( Ar 1 + , Rb ),	( P 2 + , Tl ),	( In 2 + , Pb ),
( Ti 2 + , Rb ),	( V 2 + , Tl ),	( Te 2 + , Pb ),
( Ga 2 + , I ),	( Tc 2 + , Au ),	( Al 2 + , Pb ),
( Se 2 + , I ),	( Sn 2 + , Au ),	( V 2 + , Pb ),
( Rh 2 + , I ),	( Tl 2 + , Au ),	( Tc 2 + , Po ),
( Sn 2 + , I ),	( N 1 + , Au ),	( Tl 2 + , Po ),
( P 2 + , I ),	( P 2 + , Au ),	( N 1 + , Po ),
( Cr 2 + , I ),	( V 2 + , Au ),	( P 2 + , Po ),
( Fe 2 + , I ),	( As 2 + , Hg ),	( V 2 + , Po ),
( As 2 + , Cs ),	( Tc 2 + , Hg ),	( Ga 2 + , At ),
( Rb 1 + , Cs ),	( Ru 2 + , Hg ),	( Se 2 + , At ),
( Mo 2 + , Cs ),	( Tl 2 + , Hg ),	( Rh 2 + , At ),
( Ru 2 + , Cs ),	( N 1 + , Hg ),	( Sn 2 + , At ),
( In 2 + , Cs ),	( Al 2 + , Hg ),	( Tl 2 + , At ),
( Te 2 + , Cs ),	( V 2 + , Hg ),	( N 1 + , At ),
( Al 2 + , Cs ),	( As 2 + , As ),	( P 2 + , At ),
( Ar 1 + , Cs ),	( Ru 2 + , As ),	( Cr 2 + , At ),
( Ti 2 + , Cs ),	( In 2 + , As ),	( Fe 2 + , At ),
( Tc 2 + , Se ),	( Te 2 + , As ),	( As 2 + , Ge ),
( Tl 2 + , Se ),	( Al 2 + , As ),	( Tc 2 + , Ge ),
( N 1 + , Se ),	( Ar 1 + , As ),	( Ru 2 + , Ge ),
( P 2 + , Se ),	( Ti 2 + , As ),	( In 2 + , Ge ),
( V 2 + , Se ),	( As 2 + , Ce ),	( N 1 + , Ge ),
( Tc 2 + , Te ),	( Tc 2 + , Ce ),	( Al 2 + , Ge ),
( Sn 2 + , Te ),	( Ru 2 + , Ce ),	( V 2 + , Ge ),
( Tl 2 + , Te ),	( In 2 + , Ce ),	( As 2 + , Ga ),
( N 1 + , Te ),	( N 1 + , Ce ),	( Rb 1 + , Ga ),
( P 2 + , Te ),	( Al 2 + , Ce ),	( Ru 2 + , Ga ),
( V 2 + , Te ),	( V 2 + , Ce ),	( In 2 + , Ga ),
( Fe 2 + , Te ),	( As 2 + , Fr ),	( Te 2 + , Ga ),
( As 2 + , As ),	( Rb 1 + , Fr ),	( Al 2 + , Ga ),
( Ru 2 + , As ),	( Ru 2 + , Fr ),	( Ar 1 + , Ga ),

( In 2 + , As ),	( In 2 + , Fr ),	( Ti 2 + , Ga ),
( Te 2 + , As ),	( Te 2 + , Fr ),	( As 2 + , In ),
( Al 2 + , As ),	( Al 2 + , Fr ),	( Rb 1 + , In ),
( Ar 1 + , As ),	( Ar 1 + , Fr ),	( Mo 2 + , In ),
( Ti 2 + , As ),	( Ti 2 + , Fr ),	( Ru 2 + , In ),
( Tc 2 + , Sb ),	( As 2 + , Ge ),	( In 2 + , In ),
( TI 2 + , Sb ),	( Tc 2 + , Ge ),	( Te 2 + , In ),
( N 1 + , Sb ),	( Ru 2 + , Ge ),	( Al 2 + , In ),
( P 2 + , Sb ),	( In 2 + , Ge ),	( Ar 1 + , In ),
( V 2 + , Sb ),	( N 1 + , Ge ),	( Ti 2 + , In ),
( As 2 + , Bi ),	( Al 2 + , Ge ),	( As 2 + , Ag ),
( Ru 2 + , Bi ),	( V 2 + , Ge ),	( Tc 2 + , Ag ),
( In 2 + , Bi ),	( As 2 + , Sn ),	( Ru 2 + , Ag ),
( Te 2 + , Bi ),	( Tc 2 + , Sn ),	( N 1 + , Ag ),
( Al 2 + , Bi ),	( Ru 2 + , Sn ),	( Al 2 + , Ag ),
( Ar 1 + , Bi ),	( N 1 + , Sn ),	( V 2 + , Ag ),
( Tc 2 + , TI ),	( Al 2 + , Sn ),	( P 2 + , OH ),
( Sn 2 + , TI ),	( V 2 + , Sn ),	( V 2 + , OH ),
( TI 2 + , TI ),	( As 2 + , Pb ),	( Tc 2 + , SH ),
( N 1 + , TI ),	( Tc 2 + , Pb ),	( Sn 2 + , SH ),
( Ga 2 + , BF3 ),	( Rh 2 + , UF6 ),	( TI 2 + , SH ),
( Se 2 + , BF3 ),	( Sn 2 + , UF6 ),	( N 1 + , SH ),
( Tc 2 + , BF3 ),	( TI 2 + , UF6 ),	( P 2 + , SH ),
( Rh 2 + , BF3 ),	( P 2 + , UF6 ),	( V 2 + , SH ),
( Sn 2 + , BF3 ),	( Cr 2 + , UF6 ),	( Fe 2 + , SH ),
( TI 2 + , BF3 ),	( Fe 2 + , UF6 ),	( Ga 2 + , CN ),
( N 1 + , BF3 ),	( Tc 2 + , CF3 ),	( Se 2 + , CN ),
( P 2 + , BF3 ),	( TI 2 + , CF3 ),	( Rh 2 + , CN ),
( Cr 2 + , BF3 ),	( N 1 + , CF3 ),	( Sn 2 + , CN ),
( Fe 2 + , BF3 ),	( P 2 + , CF3 ),	( P 2 + , CN ),
( Se 2 + , NO2 ),	( V 2 + , CF3 ),	( K 1 + , CN ),
( Rh 2 + , NO2 ),	( As 2 + , CCl3 ),	( Cr 2 + , CN ),
( Xe 2 + , NO2 ),	( Tc 2 + , CCl3 ),	( Fe 2 + , CN ),
( Pb 2 + , NO2 ),	( Ru 2 + , CCl3 ),	( Tc 2 + , SCN ),
( K 1 + , NO2 ),	( In 2 + , CCl3 ),	( Sn 2 + , SCN ),
( Cr 2 + , NO2 ),	( N 1 + , CCl3 ),	( TI 2 + , SCN ),
( As 2 + , O2 ),	( Al 2 + , CCl3 ),	( N 1 + , SCN ),
( Rb 1 + , O2 ),	( V 2 + , CCl3 ),	( P 2 + , SCN ),
( Ru 2 + , O2 ),	( Ga 2 + , SiF3 ),	( V 2 + , SCN ),
( In 2 + , O2 ),	( Se 2 + , SiF3 ),	( Fe 2 + , SCN ),
( Te 2 + , O2 ),	( Rh 2 + , SiF3 ),	( Ga 2 + , SeCN ),
( Al 2 + , O2 ),	( Sn 2 + , SiF3 ),	( Se 2 + , SeCN ),
( Ar 1 + , O2 ),	( P 2 + , SiF3 ),	( Tc 2 + , SeCN ),
( Ti 2 + , O2 ),	( K 1 + , SiF3 ),	( Rh 2 + , SeCN ),
( As 2 + , SF6 ),	( Cr 2 + , SiF3 ),	( Sn 2 + , SeCN ),
( Tc 2 + , SF6 ),	( Fe 2 + , SiF3 ),	( TI 2 + , SeCN ),
( Ru 2 + , SF6 ),	( As 2 + , NH2 ),	( N 1 + , SeCN ),
( TI 2 + , SF6 ),	( Tc 2 + , NH2 ),	( P 2 + , SeCN ),
( N 1 + , SF6 ),	( Ru 2 + , NH2 ),	( Cr 2 + , SeCN ),

( Al 2 + , SF6 ), ( In 2 + , NH2 ), ( Fe 2 + , SeCN ),  
 ( V 2 + , SF6 ), ( Te 2 + , NH2 ), ( TI 2 + , PH 2 ),  
 ( Ga 2 + , WF6 ), ( N 1 + , NH2 ), ( N 1 + , PH 2 ),  
 ( Se 2 + , WF6 ), ( Al 2 + , NH2 ), ( Al 2 + , PH 2 ),  
 ( Tc 2 + , WF6 ), ( V 2 + , NH2 ), ( V 2 + , PH 2 ),  
 ( Rh 2 + , WF6 ), ( Tc 2 + , PH 2 ), ( Tc 2 + , OH ),  
 ( Sn 2 + , WF6 ), ( Ru 2 + , PH 2 ), ( TI 2 + , OH ),  
 ( TI 2 + , WF6 ), ( Fe 2 + , WF6 ), ( N 1 + , OH ),  
 ( N 1 + , WF6 ), ( Ga 2 + , UF6 ), ( Cr 2 + , WF6 ),and  
 (Se 2 + , UF6 ), ( P 2 + , WF6 ).

212. (Currently Amended) In ~~a~~ A rocket comprising a walled vessel defining an interior volume containing a rocket fuel and means to direct thrust generated by said rocket fuel, wherein the improvement comprising:  
 a rocket fuel comprising a source of at least one hydrino hydride ion and a source of protons, and a means to initiate a reaction between said hydrino hydride ion and protons.
213. (Previously Presented) A rocket according to claim 212, wherein said rocket fuel further comprises at least one burn modulator.
214. (Previously Presented) A rocket according to claim 212, wherein said means for initiating said reaction between said hydrino hydride ion and said protons comprises a conventional rocket fuel.
215. (Previously Presented) A rocket according to claim 212, wherein said rocket fuel is liquid at ambient temperature and pressure.
216. (Previously Presented) A rocket according to claim 212, wherein said rocket fuel is gaseous at ambient temperature and pressure.
217. (Previously Presented) A rocket according to claim 212, wherein said rocket fuel is

solid at ambient temperature and pressure.

218. (Previously Presented) A rocket according to claim 212, wherein said source of said protons comprises an acid.
219. (Previously Presented) A rocket according to claim 218, wherein said acid is a super-acid.
220. (Previously Presented) A rocket according to claim 218, wherein said acid is selected from the group consisting of HF, HCl, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>, the reaction products of HF and SbF<sub>5</sub>, the reaction products of HCl and Al<sub>2</sub>Cl<sub>6</sub>, the reaction products of H<sub>2</sub>SO<sub>3</sub>F and SbF<sub>5</sub>, the reaction products of H<sub>2</sub>SO<sub>4</sub> and SO<sub>2</sub>, and combinations thereof.
221. (Previously Presented) A rocket according to claim 212, wherein said source of protons comprises H<sup>1</sup>.
222. (Previously Presented) A rocket according to claim 212, wherein said source of protons comprises H<sup>2</sup>.
223. (Previously Presented) A rocket according to claim 212, wherein said source of protons comprises H<sup>3</sup>.
224. (Previously Presented) A rocket according to claim 212, wherein said source of hydride ion comprises at least one compound comprising a hydrino hydride ion and at least one other element.
225. (Previously Presented) A rocket according to claim 224, wherein said compound comprises at least one hydrino atom having a binding energy of about 13.6/n<sup>2</sup> eV,

wherein  $n$  is a fraction whose numerator is 1 and denominator is an integer greater than 1 and at least one other element.

226. (Previously Presented) A rocket according to claim 224, wherein said compound comprises at least one dihydrino molecule having a first binding energy of about  $15.5/n^2$  eV, wherein  $n$  is a fraction whose numerator is 1 and denominator is an integer greater than 1 and at least one other element.
227. (Previously Presented) A rocket according to claim 224, wherein said compound comprises at least one dihydrino molecular ion having a first binding energy of about  $16.4/n^2$  eV, wherein  $n$  is a fraction whose numerator is 1 and denominator is an integer greater than 1, and at least one other element.
228. (Previously Presented) A rocket according to claim 224, wherein said compound comprises a hydrino hydride ion having a binding energy of about 0.65 eV and at least one other element.
229. (Previously Presented) A rocket according to claim 224, wherein the compound further comprises one or more selected from the group consisting of ordinary hydrogen molecules, ordinary hydride ions, ordinary hydrogen atoms, protons, ordinary hydrogen molecular ions, and ordinary  $H^{3+}$  ions.
230. (Previously Presented) A rocket according to claim 224, wherein the compound has a formula selected from the group of formulae consisting of  $MH$ ,  $MH_2$ , and  $M_2H_2$  wherein  $M$  is an alkali cation and  $H$  is selected from the group consisting of increased binding energy hydride ions, hydrino atoms and dihydrino molecules.
231. (Previously Presented) A rocket according to claim 224, wherein the compound has



the formula  $MH_n$  wherein  $n$  is 1 or 2,  $M$  is an alkaline earth cation and  $H$  is selected from the group consisting of hydrino hydride ions and hydrino atoms.

232. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MHX$  wherein  $M$  is an alkali cation,  $X$  is one of a neutral atom, a molecule, or a singly negatively charged anion, and  $H$  is elected from the group consisting of hydrino hydride ions and hydrino atoms.
233. (Previously Presented) A rocket according to claim 232, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
234. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MHX$  wherein  $M$  is an alkaline earth cation,  $X$  is a singly negatively charged anion, and  $H$  is selected from the group consisting of hydrino hydride ions and hydrino atoms.
235. (Previously Presented) A rocket according to claim 234, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
236. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MHX$  wherein  $M$  is an alkaline earth cation,  $X$  is a doubly negatively charged anion, and  $H$  is a hydrino atom.
237. (Previously Presented) A rocket according to claim 236, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.

238. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2HX$  wherein M is an alkali cation, X is a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
239. (Previously Presented) A rocket according to claim 238, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
240. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MH_n$  wherein n is an integer from 1 to 5, M is an alkali cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
241. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2H_n$  wherein n is an integer from 1 to 4, M is an alkaline earth cation and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
242. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2XH_n$  wherein n is an integer from 1 to 3, M is an alkaline earth cation, X is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
243. (Previously Presented) A rocket according to claim 242, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.

244. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2X_2H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkaline earth cation,  $X$  is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
245. (Previously Presented) A rocket according to claim 244, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
246. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2X_3H$  wherein  $M$  is an alkaline earth cation,  $X$  is a singly negatively charged anion, and  $H$  is selected from the group consisting of hydrino hydride ions and hydrino atoms.
247. (Previously Presented) A rocket according to claim 246, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
248. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2XH_n$  wherein  $n$  is 1 or 2,  $M$  is an alkaline earth cation,  $X$  is a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
249. (Previously Presented) A rocket according to claim 248, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
250. (Previously Presented) A rocket according to claim 224, wherein the compound has

the formula  $M_2XX'H$  wherein M is an alkaline earth cation, X is a singly negatively charged anion, X' is a doubly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.

251. (Previously Presented) A rocket according to claim 250, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
252. (Previously Presented) A rocket according to claim 250, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
253. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MM'H_n$  wherein n is an integer from 1 to 3, M is an alkaline earth cation, M' is an alkali metal cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
254. (Previously Presented) A rocket according to claim 224, wherein the compound is  $MM'XH_n$  wherein n is 1 to 2, M is an alkaline earth cation, M' is an alkali metal cation, X is a singly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
255. (Previously Presented) A rocket according to claim 254, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
256. (Previously Presented) A rocket according to claim 224, wherein the compound is  $MM'XH$  where M is an alkaline earth cation, M' is an alkali metal cation, X is a doubly

negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.

257. (Previously Presented) A rocket according to claim 256, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
258. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MM'XX'H$  where M is an alkaline earth cation, M' is an alkali metal cation, X and X' are each a singly negatively charged anion, and H is selected from the group consisting of hydrino hydride ions and hydrino atoms.
259. (Previously Presented) A rocket according to claim 258, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
260. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $H_nS$  wherein n is 1 or 2, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species.
261. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MSiH_n$  wherein n is an integer from 1 to 6, M is an alkali or alkaline earth cation, and the hydrogen content of  $H_n$  comprises at least one increased binding energy hydrogen species.
262. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MXM'H_n$  wherein n is an integer from 1 to 5;  
M is an alkali or alkaline earth cation;

X is a singly negatively charged anion or a doubly negatively charged anion;

M' is selected from the group consisting of Si, Al, Ni, the transition elements, the inner transition elements, and the rare earth elements; and

the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

263. (Previously Presented) A rocket according to claim 262, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
264. (Previously Presented) A rocket according to claim 262, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
265. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MAIH_n$  wherein n is an integer from 1 to 6, M is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
266. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MH_n$  wherein:
- n is an integer from 1 to 6;
  - M is selected from the group consisting of the transition elements, the inner transition elements, and the rare earth element cations and nickel; and
  - the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
267. (Previously Presented) A rocket according to claim 224, wherein the compound has

the formula  $MNiH_n$  wherein:

n is an integer from 1 to 6;

M is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum; and

the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

268. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MM'H_n$  wherein:

n is an integer from 1 to 6;

M is selected from the group consisting of alkali cations, alkaline earth cations, silicon, and aluminum;

M' is selected from the group consisting of the transition elements, the inner transition elements, and rare earth element cations; and

the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

269. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $M_2SiH_n$  wherein n is an integer from 1 to 8, M is an alkali or alkaline earth cation, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

270. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $Si_2H_n$  wherein n is an integer from 1 to 8, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.

271. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $SiH_n$  wherein n is an integer from 1 to 8, and the hydrogen content  $H_n$

comprises at least one increased binding energy hydrogen species.

272. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $TiH_n$  wherein  $n$  is an integer from 1 to 4, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
273. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $Al_2H_n$  wherein  $n$  is an integer from 1 to 4 and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
274. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MXAlX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy hydrogen species.
275. (Previously Presented) A rocket according to claim 274, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
276. (Previously Presented) A rocket according to claim 274, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
277. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $MXSiX'H_n$  wherein  $n$  is 1 or 2,  $M$  is an alkali or alkaline earth cation,  $X$  and  $X'$  are each either a singly negatively charged anion or a doubly negatively charged anion, and the hydrogen content  $H_n$  comprises at least one increased binding energy



hydrogen species.

278. (Previously Presented) A rocket according to claim 277, wherein said singly negatively charged anion is selected from the group consisting of halogen ions, hydroxide ions, hydrogen carbonate ions, and nitrate ions.
279. (Previously Presented) A rocket according to claim 277, wherein said doubly negatively charged anion is selected from the group consisting of carbonate ions and sulfate ions.
280. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $\text{SiO}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6 and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species.
281. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $\text{MSiO}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species.
282. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $\text{MSi}_2\text{H}_n$  wherein  $n$  is an integer from 1 to 6,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species.
283. (Previously Presented) A rocket according to claim 224, wherein the compound has the formula  $\text{M}_2\text{SiH}_n$  wherein  $n$  is an integer from 1 to 8,  $M$  is an alkali or alkaline earth cation, and the hydrogen content  $\text{H}_n$  comprises at least one increased binding energy hydrogen species.

284. (Previously Presented) A rocket according to claim 224, wherein the compound is greater than 50 atomic percent pure.
285. (Previously Presented) A rocket according to claim 224, wherein the compound is greater than 90 atomic percent pure.
286. (Previously Presented) A rocket according to claim 224, wherein said at least one other element comprises at least one selected from the group consisting of a proton, ordinary hydrogen atom, ordinary hydride ions, ordinary hydrogen molecules, ordinary hydrogen molecular ions and ordinary  $H_3^+$  ion.
287. (Previously Presented) A rocket according to claim 224, wherein said at least one other element comprises at least one element selected from the group consisting of alkaline earth metals and alkali metals.
288. (Previously Presented) A rocket according to claim 287, wherein said element comprises lithium or lithium ion.
289. (Previously Presented) A rocket according to claim 224, wherein said at least one other element comprises at least one element selected from the group consisting of organic compounds.
290. (Previously Presented) A rocket according to claim 224, wherein said at least one other element comprises at least one element selected from the group consisting of semiconductors .
291. (Previously Presented) A rocket according to claim 224, wherein said compound

comprising:

(a) at least one neutral, positive or negative increased binding energy hydrogen species having a binding energy:

- (i) greater than the binding energy of the corresponding ordinary hydrogen species, or
- (ii) greater than the binding energy of any hydrogen species for which the corresponding ordinary hydrogen species is unstable or is not observed because the ordinary hydrogen species' binding energy is less than thermal energies at ambient conditions, or is negative; and

(b) at least one other element, wherein said increased binding energy hydrogen species is selected from the group consisting of  $H_n$ ,  $H_n^-$ , and  $H_n^+$ , where n is an integer of 1 to 8, and n is greater than 1 when H has a positive charge.

292. (Previously Presented) A rocket according to claim 291, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydride ion having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where p is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and e is the elementary charge; (b) hydrogen atom having a binding energy greater than about 13.6 eV; (c) hydrogen molecule having a first

binding energy greater than about 15.5 eV; and (d) molecular hydrogen ion having a binding energy greater than about 16.4 eV.

293. (Previously Presented) A rocket according to claim 291, wherein the increased binding energy hydrogen species comprises a hydride ion having a binding energy of about 3.0, 6.6, 11.2, 16.7, 22.8, 29.3, 36.1, 42.8, 49.4, 55.5, 61.0, 65.6, 69.2, 71.53, 72.4, 71.54, 68.8, 64.0, 56.8, 47.1, 34.6, or 19.2.

294. (Previously Presented) A rocket according to claim 291, wherein said increased binding energy hydrogen species is a hydride ion having a binding energy greater than the binding energy of the corresponding ordinary hydride ion for  $p = 2$  up to 23 in which the binding energy is represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi \mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^3} \right)$$

where  $p$  is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge.

295. (Previously Presented) A rocket according to claim 291, wherein said increased binding energy hydrogen species is selected from the group consisting of (a) a hydrino atom having a binding energy of about  $13.6 \text{ eV}/(1/p)^2$ , where  $p$  is an integer greater than 1; (b) a hydride ion having a binding energy represented by

$$\text{Binding Energy} = \frac{\hbar^2 \sqrt{s(s+1)}}{8\mu_e a_0^2 \left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^2} - \frac{\pi\mu_0 e^2 \hbar^2}{m_e^2 a_0^3} \left( 1 + \frac{2^2}{\left[ \frac{1 + \sqrt{s(s+1)}}{p} \right]^5} \right)$$

where  $p$  is an integer greater than 1,  $s = \frac{1}{2}$ ,  $\hbar$  is Plank's constant bar,  $\mu_0$  is the permeability of vacuum,  $m_e$  is the mass of the electron,  $\mu_e$  is the reduced electron mass,  $a_0$  is the Bohr radius, and  $e$  is the elementary charge; (c) a trihydrino molecular ion,  $H_3^+$  ( $1/p$ ), having a binding energy of about  $22.6/(1/p)^2$  eV; (d) a dihydrino molecule having a binding energy of about  $15.5/(1/p)^2$  eV; and (e) a dihydrino molecular ion with a binding energy of about  $16.4/(1/p)^2$  eV.

296. (Previously Presented) A rocket according to claim 295, wherein  $p$  is 2 to 200.
297. (Previously Presented) A rocket according to claim 295, wherein  $p$  is 2 to 24.
298. (Previously Presented) A rocket according to claim 295, wherein said increased binding energy hydrogen species is negative.
299. (Previously Presented) A rocket according to claim 212, wherein said source of hydrino hydride ion comprises a source of 2 or more types of hydrino hydride ions.
300. (Presented Previously) A compound having the formula  $[KH_m KCO_3]_n$  wherein at least one H is a hydrino atom having a binding energy of about  $13.6/n^2$  eV,  $n$  is a fraction in which the numerator is 1 and denominator an integer greater than one.
301. (New) A method according to claim 105, comprising rapid thermal decomposition of the hydrino hydride.